

In The Matter Of:
USA, et al. v.
E.I. du Pont de Nemours and Company

Luis Chu
December 06, 2013

Wilcox & Fetzer, Ltd.
1330 King Street
Wilmington, DE 19801
email: depos@wilfet.com, web: www.wilfet.com
phone: 302-655-0477, fax: 302-655-0497



IN THE UNITED STATES DISTRICT COURT
MIDDLE DISTRICT OF LOUISIANA

UNITED STATES OF AMERICA,)	
ET AL, JEFFREY M. SIMONEAUX,)	
Realtor,)	
)	Civil Action
Plaintiffs,)	No. 3:12-cv-219
)	
v)	
)	
E.I. du PONT de NEMOURS AND)	
COMPANY,)	
)	
Defendant.)	

Deposition of LUIS CHU taken
pursuant to notice at the law offices of Potter,
Anderson & Corroon, 1313 North Market Street, 6th
Floor, Wilmington, Delaware, beginning at 10:19
a.m. on Friday, December 6, 2013, before
Christina M. Vitale, Certified Court Reporter and
Notary Public.

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1 APPEARANCES:

2 JANE H. BARNEY, ESQUIRE
3 J. H. BARNEY LAW FIRM, LLC
4 2561 CitiPlace Court, Suite 750-161
5 Baton Rouge, Louisiana 70808
6 For the Plaintiffs

7 MONIQUE M. WEINER, ESQUIRE
8 KUCHLER POLK SCHELL & RICHESON, LLC
9 1615 Poydras Street, Suite 1300
10 New Orleans, Louisiana 70112
11 For the Defendant

12 - - - - -

13 LUIS CHU,

14 the deponent herein, having first
15 been duly sworn on oath, was
16 examined and testified as follows:

17 EXAMINATION

18 BY MS. BARNEY:

19 Q. Good morning, Mr. Chu.

20 A. Good morning.

21 Q. Could you state your full name and
22 address for the record, please.

23 A. Home address?

24 Q. Yes.

25 A. Luis Chu, 1642 Flint Hill Road,
26 Landenberg, P. A., 19350.

27 Q. How long have you lived there?

28 A. Since February of 2004.



1 Q. Have you ever given a deposition before?

2 A. No.

3 Q. Like this?

4 A. Not at all. Not even like this.

5 Q. I'll just tell you a little bit about the
6 way it works.

7 A. Okay.

8 Q. I'm going to ask you questions and you
9 can answer those questions. The court reporter
10 will write down what we each say. So, one thing
11 we need to watch for is talking over each other.
12 And I'm really bad about that too.

13 A. Yeah.

14 Q. You sort of get in a conversation and the
15 other one knows where the other one is going and
16 you kind of start stepping on each other. If we
17 can try to avoid that, it helps the court
18 reporter.

19 A. Okay.

20 Q. If you can say yes or no instead of
21 saying mm-hmm --

22 A. Okay.

23 Q. -- because even though the court reporter
24 can tell if it's yes or no --



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1 A. Okay is not -- okay.

2 MS. WEINER: You can say okay as
3 well.

4 Q. Try to say yes or no and we'll try to
5 remind you if we get off on that. We'll take a
6 break whenever you need one and then we'll have
7 lunch brought in; and, if you need a break, just
8 say you need a break and we can stop for a little
9 while.

10 A. Okay.

11 Q. If there is any question that you don't
12 understand, just ask me to repeat it or rephrase
13 it and I will do that.

14 A. Okay.

15 Q. If you answer it, then I'll assume you
16 understood the question, but, if you need any
17 clarification, just tell me. I'm assuming you
18 are not on any medication that would affect the
19 way you can process questions and answer them
20 today. Is there anything that is impairing your
21 ability to hear questions and give answers?

22 A. No. I'm not under any kind of
23 medication. I was waiting for you to finish
24 talking.



1 Q. Thank you. That's just a standard
2 question we ask.

3 A. Okay.

4 Q. Can you tell me I guess we'll start with
5 what you reviewed to prepare for your deposition
6 today? We had a little bit of a conversation off
7 the record where there were two documents that
8 you looked at.

9 A. Right. I mean, it wasn't so much a
10 preparation for the deposition. I told Monique
11 that I have regular conversations with Percy
12 Bell, he is a good friend of mine, and, so, I was
13 aware that the depositions have been going on for
14 a while. So, when I told him that I've been
15 called to make a deposition he was talking, Oh,
16 yeah, there has been some discussion about gas
17 leaks. And I told him, Yeah, I never been
18 involved in any of the incident investigations,
19 but I recall checking a calculation because it's
20 a form -- you have probably seen it. It's a form
21 for the gas leaks and I want to take a look at it
22 and see when I reviewed that.

23 Of course, I didn't realize that
24 until yesterday that once I looked at that it



1 becomes part of a preparation even though in my
2 mind it was not technically review. I just
3 wanted to see the time frame of when I looked at
4 it.

5 Q. And that's fine. I think Monique is just
6 probably trying to prepare you so you are not
7 caught off guard when I ask you those kind of
8 questions.

9 A. Yeah, yes, and she did a very good job --
10 she said, Let's just print it out and have it
11 available because it's a few pages and I said,
12 Sure, but I just forgot to bring it.

13 Q. I appreciate you are having somebody
14 bring those to us today and we'll get to that
15 maybe after lunch or right before. Other than
16 maybe some conversations -- let me strike that.
17 You didn't really do anything to prepare
18 specifically for the deposition other than look
19 at those documents to the extent that was
20 preparation?

21 A. Other than spending a couple hours
22 yesterday to go over some of the things that you
23 just mentioned as far as not talking over each
24 other and things like that, no.



1 Q. That was a meeting between you and Ms.
2 Weiner, right?

3 A. Yes.

4 Q. And I think you are representing Mr. Chu?

5 MS. WEINER: That's correct.

6 Q. I don't want to ask you anything about
7 what you all talked about because that's
8 privileged. Other than that meeting and looking
9 at those documents is there anything else that
10 you did?

11 A. No.

12 Q. No research or legal research or anything
13 like that?

14 A. No. Again, this is the first time I
15 heard about depositions before. It was the first
16 time I experienced one, but, no, I didn't. I've
17 been traveling so much in the last couple months
18 that I haven't even looked at it.

19 Q. Thanks for coming in today, by the way, I
20 appreciate you meeting with us. We didn't have
21 to subpoena you and DuPont agreed to make you
22 available, so, I appreciate it. Why don't we
23 start with your educational background. Where
24 did you live when you graduated from high school?



1 A. I lived in Los Angeles. I went to Santa
2 Monica High School in California. Lived in Los
3 Angeles and went to University of California, Los
4 Angeles, for undergraduate. Then, worked a
5 couple years before going to grad school in
6 Golden, Colorado. School's name, Colorado School
7 of Mines.

8 Q. How do you spell that, mines, M-I-N-E-S?

9 A. Yes, mines. Both my degrees are in
10 chemical engineering.

11 Q. You have a B. S. from the University of
12 California in chemical engineering?

13 A. Right, and M. S. from Colorado School of
14 Mines. Bachelor's from '81 and Master's in '86.

15 Q. Where did you work in between undergrad
16 and grad school?

17 A. Consulting company called Global
18 Geochemistry.

19 Q. What did you do for them during that time
20 period?

21 A. I provided designs for analytical
22 instruments. The company was owned by a
23 professor at UCLA where I worked as -- in his lab
24 for maybe three years. So, when I -- before I



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1 graduated he offered me a job.

2 Q. Sounds like a compliment.

3 A. Well, actually helped me quite a bit
4 because he is the one to recommended that I look
5 at Colorado School of Mines, which I never heard
6 of, and turned out to be a very good school,
7 location. Still have a lot of good friends.

8 Q. That's great. So, when you got your
9 chemical engineering degree -- let me go back,
10 I'm sorry, go back to that job at Global.

11 A. Mm-hmm.

12 Q. Was there a particular industry that you
13 were designing analytical instruments for?

14 A. This was primarily for oil and gas
15 industries. Some of the analytical instruments
16 or pretty much all the analytical instruments had
17 to do with exploration of oil and gas. So,
18 locating reservoirs of oil and gas.

19 Q. Did you have any particular emphasis in
20 undergrad or was that just a general chemical
21 engineering degree?

22 A. Well, the emphasis is chemical
23 engineering. The overall I'll say it would be
24 engineering. You know, typical bachelor's degree



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1 the first two, two and a half years, almost all
2 engineers take. Then, after that it narrows into
3 specialties, chemical, electrical.

4 Q. And when you were doing your master's in
5 chemical engineering, is there any breakdown
6 where you kind of focus on particular things or
7 is it still just a general chemical engineering?

8 A. Well, it's still chemical engineering.
9 For a graduate program usually you pick a
10 project, a thesis, and if I were to narrow it
11 down, I chose experimental. So, running
12 experiments to get data to arrive at the thesis
13 to report as opposed to some of my colleagues had
14 computer simulations that didn't involve no
15 actual lab work; but, the graduate programs
16 typical for engineering is -- especially master's
17 is not to focus on a thesis and take that and be
18 the lifelong specialty.

19 It's just as somebody put to me it's
20 to show how much you can develop on your own, how
21 much you can be self-motivated as opposed to a
22 bachelor's program, which they teach you how to
23 learn; and, a graduate program is, well, how much
24 can you teach yourself, yes.



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1 Q. Did your thesis have a title, one that we
2 could understand?

3 A. Well, tell you the truth I don't remember
4 the title exactly, but it had to do with
5 supercritical fluids. To put in perspective it's
6 a study to if you subject gases to high enough
7 pressure and temperature they start behaving more
8 like liquids. So, my thesis was to look at using
9 a gas in the natural state, which is CO₂, and
10 apply enough pressure and temperature so it will
11 behave as a liquid and then act as a solvent to
12 do selective extractions from a variety of
13 compounds.

14 Q. That sounds kind of relevant to your work
15 at DuPont, which we'll get to later, but it
16 sounds in the ballpark anyway.

17 A. Well, pressures for supercritical fluids
18 are probably three, three orders of magnitude.
19 They're very high pressures. The work that is
20 related to sulfuric acid will work on much lower
21 pressures, but at much higher temperatures, yes.

22 Q. After you got your master's degree where
23 did you go?

24 A. Worked for an engineering consulting



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1 company in Irvine, California called Fluor
2 Daniel.

3 Q. Fluor Daniel?

4 A. Yes, F-L-U-O-R, and Daniel, yes.

5 Q. What job did you do for them when you got
6 there?

7 A. As a process engineer working in
8 evaluations of coal gasification. Essentially
9 taking a solid piece of coal and converting it
10 into a combustible gas that can then be used to
11 power turbines to generate electricity and then
12 the exhaust from the turbines can be used to heat
13 water to generate steam, to drive turbines to
14 generate more electricity.

15 So, it's -- probably two and a half
16 years I worked in the company this was just
17 primarily I'll call it paper studies where we do
18 engineering and cost estimates of what will it
19 take to take a material like coal, taking it to a
20 gaseous fuel and for utility companies to use it
21 to generate electricity and compare the costs,
22 you know, the costs to just hydroelectric,
23 nuclear, burning natural gas, burning fuel.

24 Q. How long did you work at Fluor Daniel?



1 A. I think about two and a half years.

2 Q. And what prompted you to leave that job?

3 A. Because it was just paper studies. It
4 was during the time that oil and gas were very
5 cheap. So, it was not cost-effective to convert.
6 Even though it was a very neat technology two and
7 a half years of paper studies at least for
8 engineer like myself is not idea of fun.

9 Q. Where did you go after that?

10 A. All right, so, I worked for an
11 environmental company that provided consulting
12 services to refineries in chemical industries.

13 Q. What was the name of that company?

14 A. Mittel, M-I-T-T-E-L, H-A-U-S-E-R, it's
15 one word. I don't think they're in business
16 anymore.

17 Q. About how many employees did they have?
18 Was it a small firm?

19 A. It was a small firm. They had I think
20 four branches. I was in the Laguna Hills,
21 California branch and we probably had a couple
22 dozen and I think overall the company had 100,
23 I'm not sure.

24 Q. And what was your job at Mittelhauser?



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1 A. At that time refineries had to comply
2 with TCLP regulations -- toxicity,
3 characteristic, leaching, procedure, I think
4 that's the acronym, it has been a few years --
5 but essentially the refineries had to comply with
6 certain wastewater standards. For example,
7 benzene is a component that finds its way into
8 refinery wastewater. My job was to try to work
9 with refineries to try to isolate, identify, the
10 sources where the benzene was originating and how
11 to minimize it so that they can comply with the
12 wastewater discharge.

13 Other projects that I worked on
14 related to hazardous waste handling, particularly
15 from refineries. There are some -- in the
16 regulations there are some listed ways. For
17 example, refineries, if I recall correctly,
18 they're what they are called K listed. K has to
19 do with the regulation subtopic, but the
20 consulting company that I used to work for looked
21 at ways of how to handle those hazardous wastes,
22 whether that be through incineration or through
23 solidification, in such a way that water would
24 not leach contaminants out.



1 Q. So, you were -- sounds like you were
2 working in the process part of that --

3 A. Right, right. Pretty much I would say
4 all my jobs that I had had to relate with the
5 process. We had -- for example, Mittelhauser
6 there were other professionals like geologists,
7 hydrogeologists. Half of us I think were
8 chemical engineers.

9 Q. So, your title was chemical engineer or
10 process engineer at Mittelhauser or were there
11 not really titles?

12 A. Well, title sometimes doesn't mean --
13 I'll be like senior engineer and you just had to
14 do with number of years of experience or
15 sometimes the company tell us to have a title on
16 our business card just to project to the
17 customer, okay --

18 Q. What you are?

19 A. Right.

20 Q. Who you are?

21 A. Right.

22 Q. So, in the job with Mittelhauser was it
23 your job to work on the process -- process issues
24 to -- well, let me strike that and start over.



1 Was it your job at all to decide which
2 environmental statute was at issue for a project
3 or was it just your job to look at the process
4 once somebody said, This is what we need to meet
5 environmentally, can you help us do that? If
6 that makes sense.

7 A. Yeah. No, my role was not to determine
8 whether they were or were not in compliance. My
9 role was already identified even internally by
10 the customer that either they were not going to
11 be in compliance or they were not sure, right?
12 So, I would be asked, for example, one project
13 that I worked for for a client was to do a
14 complete evaluation and sources of wastewater
15 from the whole refinery that makes it way to the
16 wastewater treatment facility.

17 The purpose of doing that was to, you
18 know, first, to identify that, for example,
19 benzene was making its way to the wastewater
20 facility, but because you can imagine the size of
21 a refinery and so many drains and taps it was
22 trying to go unit by unit in each of the
23 refineries and interview operators, look at the
24 process diagrams, confirm that whether the number



1 of sources that the operator thought were
2 discharging into the wastewater really were in
3 agreement with P & ID, which is process and
4 instrument diagrams, N is the ampersand.

5 So, I guess in the second part that
6 you meant which is my role is -- I don't know --
7 they don't know where the source is coming from
8 so you just have to map it. I mean, it was a big
9 three-month project. I remember the report was
10 probably three inches just to show, you know,
11 area per area, which sources, an estimate of how
12 much material is being discharged. Ultimately,
13 we did find the primary sources and were able to
14 segregate those so that the wastewater facility
15 did not have to put additional treatment facility
16 to address the overall volume of the wastewater.

17 Q. So, you were sort of figuring out from a
18 process standpoint what chemicals or waste was
19 being generated so that the people who were
20 trying to comply with the statute could know what
21 they were dealing with --

22 A. Right.

23 Q. -- and try to address it?

24 A. Yes. In this case -- my apologies.



1 Q. No, that's okay.

2 A. But in this case they originally thought
3 if they didn't identify the sources they had to
4 treat the several thousand gallons per minute of
5 wastewater by identifying where the major two
6 sources are and handling those properly they
7 didn't have to put in additional equipment. They
8 were already in compliance. We put in a test
9 program for them, a test program for that, to be
10 able to verify and sustain that type of low
11 concentration discharge.

12 Q. Do you recall which refineries you did
13 that work at?

14 A. I do, but I'm not sure if --

15 Q. You think it might be confidential, okay.
16 That's fine.

17 A. I'm not sure.

18 Q. That's okay, if you don't feel
19 comfortable it's not pertinent enough to our
20 discussion to make you feel like you might be
21 breaching that.

22 A. I mean, this has been over 20 years ago.
23 I know the refinery, but, again, it's not a
24 customer -- if it's not relevant --



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1 Q. It's not that relevant so I'll leave that
2 one alone.

3 A. Again, I was proud of that job because
4 sometimes it's not -- when it's needed, you know,
5 we want the company to put in a treatment
6 facility, but it's best if you can segregate it
7 in a very small amount and change their practices
8 and that is better than trying to treat
9 uncontrolled discharge.

10 Q. That felt good, you were successful in
11 that effort?

12 A. Yeah, the client was happy too.

13 Q. It sounds like you just had less water to
14 treat once you identified the source and sort of
15 steered it back --

16 A. Significantly less.

17 Q. How long were you with Mittelhauser?

18 A. I think probably only about a year and a
19 half. It was less than two years.

20 Q. And what caused you to leave there?

21 A. Well, I had some friends that I went to
22 school with and they worked at a research
23 facility for Conoco, which is an old company that
24 has since been acquired by Phillips Petroleum.



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1 They knew that I wanted to go into an R&D
2 organization and Conoco was one of the few
3 companies that don't look at graduate degrees.
4 Most R&D organizations, especially oil companies,
5 would not have hired anybody unless they had a
6 Ph.D., but I knew some friends at Conoco that
7 recommended me.

8 The person who eventually became my
9 supervisor he used to be a professor at Colorado
10 School of Mines. So, he had gotten some
11 recommendations from my other professors, my
12 advisor. So, he actually worked -- it actually
13 worked out okay. In fact, before I took the job
14 at Mittelhauser I was offered a job by Conoco,
15 but I just got married and Conoco R&D was located
16 in Ponca City, Oklahoma. So, it's a city of
17 20,000 and I wasn't sure I was going to be able
18 to convince my wife to move from Los Angeles to a
19 town of 20,000.

20 Q. That was probably wise on your part. And
21 you are still married?

22 A. Well, again, still married. So, the
23 first offer I declined and took the job at
24 Mittelhauser. Then, on second thoughts the costs



1 of living in Los Angeles are significantly higher
2 than Oklahoma. So, a number of reasons. You
3 asked me why I moved, you know, the job was
4 right, I knew people there. We would not have to
5 depend on two incomes to survive. In Los Angeles
6 two incomes might not be enough.

7 Q. How long were you at Conoco?

8 A. Technically at that time Conoco was owned
9 by DuPont. So, I was at Conoco four years. So,
10 as far as -- I forget if you asked me the years
11 of service. I've been with DuPont about
12 24 years. So, four of those years were with
13 Conoco, but since it was owned by DuPont it's
14 part of my service.

15 Q. About what years was that?

16 A. 1990 to '94.

17 Q. And just sort of generally what did you
18 do in the R&D department at Conoco?

19 A. There were a number of projects related
20 to the production of oil and gas.

21 Q. And then what change did you make in
22 1994?

23 A. I transferred to DuPont proper. I
24 transferred to DuPont Engineering.



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1 Q. What prompted you to do that?

2 A. Well, Conoco I was there four years and
3 at that time it's very tough for the oil
4 industry. So, there were three reorganizations
5 in four years. Our division probably lost half,
6 if not more, of the staff in three years and my
7 supervisor -- I said Conoco at that time was a
8 subsidiary wholly-owned by DuPont; but, our
9 e-mail system at the time, which was very basic,
10 we did not have a lot of interaction with DuPont,
11 but my supervisor knew of a job within DuPont
12 that would fit my skills. He was a die-hard
13 Conoco employee, 40-some years, and he was
14 saddened to see a lot of his people being not --
15 well, terminated in that period of time. So, he
16 suggested to look at a job at DuPont.

17 I came to Wilmington and I spoke with
18 the group. I liked the group and it turned out
19 to be a good decision.

20 Q. Good. You are obviously one of the
21 valuable folks at Conoco since no matter how many
22 cuts they did you weren't one of them, right?

23 A. No, I look at I was probably low enough
24 in the totem pole that carrying me was not enough



1 to impact the bottom line because there were a
2 lot of good people, but a lot of them took early
3 retirement. They get incentive, years of
4 service, weeks of paid vacation per year. Some
5 of them did really well.

6 Part of the reason again I left is
7 after you cut enough then you don't have the
8 manpower to do the work that our group is
9 supposed to be doing, so, it was just a matter of
10 time.

11 Q. Who did you meet with when you came to
12 Wilmington? Do you remember?

13 A. Yes. He retired, but he is a very good
14 friend of mine, Tony Pezzone.

15 Q. And he was an engineer?

16 A. Yes, chemical engineer. He was an
17 engineer supervisor for at that time the nylon
18 intermediates group.

19 Q. When you started at DuPont proper, what
20 did you start doing at that point?

21 A. Also as a chemical engineer. Our group,
22 our engineering group, provided technical support
23 for the nylon intermediates division. This is
24 the division that makes components that goes into



1 making the nylon fiber and makes a variety of
2 products from carpet to engineering products.

3 Q. How long did you work with the nylon
4 intermediates?

5 A. Until 2001. It was in two locations, in
6 Wilmington at that time, and then in Houston,
7 Texas. Our group moved to Texas, specifically to
8 Houston, which is halfway to two of duPont's
9 biggest nylon intermediates plants at the time,
10 which were in Sabine, Texas and Victoria, Texas.

11 Q. What did you start doing for DuPont in
12 2001?

13 A. I remained in engineering, but switched
14 over to biobased materials. I moved to Illinois,
15 Decatur, Illinois, where we had a joint
16 development project with a company called Tate &
17 Lyle. It was a pilot plan to try to develop what
18 we called basic data, data that we can use to
19 build a production plant. The project entailed
20 taking genetically modified bacteria, feeding
21 them sugars from corn and have them generate a
22 product called propanediol. So, essentially they
23 ingest the sugars and they create a chemical
24 called propanediol.



1 So, you can think of it as it's a
2 fermentation process and the chemical then is
3 filtered from the biomass, the water is
4 evaporated, the bi-products are distilled and you
5 end up with a very pure product. Propanediol can
6 be used as a substitute for petroleum-based
7 propanediol. So, it's a biobased product to make
8 a chemical to be used for fibers.

9 Q. How long did you work in that area with
10 biobased materials?

11 A. About 2001, to the end of 2003. So,
12 three and a half years.

13 Q. What did you start doing in 2003?

14 A. Actually, 2004. That would be like
15 January 2004 I left the DuPont engineering group
16 and start working for one of the DuPont,
17 businesses, which at that time it was called
18 DCSE, DuPont Chemicals Solutions Enterprise. The
19 reason I remember all these dates is because I
20 had to put in -- with my new group I had to put
21 in a biography.

22 Q. So, you have done it in the recent past?

23 A. But I had to condense it in a very small
24 paragraph.



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1 Q. You are doing a good job of it. Did you
2 have to relocate to start doing that?

3 A. Yes, I relocated from Illinois back to
4 the Ivory Tower.

5 Q. Wilmington? Okay. How long were you in
6 the DCSE group?

7 A. DCSE changed their names to DC&F, DuPont
8 Chemicals and Floor products. So, even though
9 it's two different divisions, still same group.
10 So, I was there until technically October of last
11 year in 2012. In reality I was -- I transferred
12 all my duties, responsibilities, probably
13 sometime in towards the end of the first quarter
14 of 2012. It was a long transition.

15 Q. When you were going through that
16 transition, who were you transitioning your
17 duties to?

18 A. George Brown, who is one of the members
19 of my previous group, ATC, Acid Technology
20 Center.

21 Q. He was with the Acid Technology Center or
22 you were too?

23 A. He is still with the Acid Technology
24 Center, yes.



1 Q. Was that part of the DCSE?

2 A. Right, DCSE think of it as a big business
3 unit. My group was called Acid Technology
4 Center, yes.

5 Q. What were your responsibilities in the
6 Acid Technology Center when you started in 2004;
7 and, then, if they changed, you can kind of tell
8 me briefly how they changed.

9 A. Then, I have to backtrack a little more
10 because I had two -- I worked for two different
11 groups within DCSE. From 2004 until 2006 I was
12 in the research part and I have duties in
13 calculation. Then, towards the last part of that
14 tour of duty I had some responsibilities for
15 sulfuric acid.

16 Q. Generally what types of calculation work
17 did you do in 2004 to 2006?

18 A. They were primarily what I call
19 processing relations, which is using a computer
20 program to simulate or to do material and energy
21 balances for calculation plans and part of it for
22 sulfuric acid plants. The purpose were to use
23 that information to roughly size equipment
24 involved in those projects and do a capital



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1 estimate. So, part of it was try to figure out
2 ways to optimize the design to make the plant at
3 lower cost and making it more cost-effective for
4 us to build and operate plants.

5 Q. When you started working with
6 responsibility for sulfuric acid in 2006, how did
7 your job change at that point?

8 A. Well, it went from just primarily, again,
9 paper studies, there were some -- at that time we
10 just -- we built the first new acid plant in I
11 don't know how long and this was in Delaware
12 City, Delaware. The plant is called Red Lion.
13 So, the plant had some start-up problems, some
14 technical issues, and my boss volunteered me to
15 go help. So, I went to the new plant, one of
16 thousands, and long story short I start
17 interacting with the plant managers, the
18 operators, and was able to contribute to some of
19 the solutions that they needed.

20 And I asked for a transfer from,
21 let's say, the R&D part of my group to another
22 group, ATC. My condition was the transfer was to
23 support that plant, which the plant manager
24 accepted and my old boss agreed and my new boss



1 agreed.

2 Q. So, you were working under -- let's see,
3 you were working under ATC from 2006 to 2012?

4 A. Right.

5 Q. But more in the R&D part?

6 A. No, no. From 2004 to 2006 it would be
7 more in the R&D part. After that, after 2006,
8 once I became part of ATC, the ATC group is
9 comprised of a number of professionals, mostly
10 chemical engineers. There were a couple chemists
11 or there is still a couple chemists and our role
12 -- the group's role was to provide kind of a
13 centralized resource for all DuPont acid plants
14 where they can go and ask for technical help,
15 maintenance, reliability, turnaround support.
16 So, that's the mission of the group.

17 The way it was organized is each acid
18 plant had one engineer or a chemist assigned to
19 that plant as the primary contact. Some people
20 in my group had two sites because they just
21 happened to be very similar or they're small
22 enough and, again, even though we were assigned
23 in my case to the Red Lion plant that didn't
24 preclude the plant management from asking



1 resources from our group or from DuPont in
2 general. So, just to make it easier for each of
3 us to get intimately familiar with the plant so
4 that if there is a question we can be able to
5 address it a lot easier.

6 Q. Now, was that the 2004 to 2006 role or
7 was that 2006 to later?

8 A. 2006 into 2012.

9 Q. You became the person sort of assigned to
10 the Red Lion plant in 2006?

11 A. Right, until 2010, sometime in the first
12 quarter.

13 Q. Did they close Red Lion at that point?

14 A. At that point they -- we idled it, which
15 means you would put it in a state that we felt
16 comfortable restarting it within a year or two.
17 The reason they idled the plant is because the
18 they closed the refinery right next to it and the
19 refinery provided essentially all the utilities
20 to the Red Lion plant.

21 Q. So, your work from 2006 to 2010 was more
22 hands-on I guess at the plant or more involved
23 with the people at the plant and the operations
24 at the plant than it had been before?



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1 A. Well, let me see how to describe. Let me
2 describe in two parts. The typical role for what
3 we call ATC engineers, my role, is we have
4 members in our group that were very hands-off,
5 meaning they will go to the site once a quarter
6 or only when the plant called them because in
7 general our group, let's say the people
8 supporting the plants, they're located in a
9 different state, okay?

10 So, it was up to the plant personnel
11 and up to the assigned engineer or chemist to
12 determine how often they visited the plant. So,
13 we had, again, there were people that I knew that
14 only went once a quarter or when needed. There
15 was another engineer that has since retired that
16 visited the plant every other week whether needed
17 or not, just part of it.

18 In my case when I start supporting
19 Red Lion from 2006, again, to 2010 the plant was
20 located or is located about the same distance
21 from my house to my office and I had a very good
22 working relationship with the two plant managers
23 that were there during my tenure and the
24 operating and maintenance staff and the



1 operators. So, it became a lot easier for me
2 even to just go to the plant initially and just
3 sit in the conference room, but they later find
4 me a office where I can just go there. If they
5 had any questions, I was there. If I hear
6 anything in the hallway that I could contribute,
7 I participated. It was a very good working
8 relationship that I rather be there than in my
9 office --

10 Q. Trying to call them and deal with it that
11 way?

12 A. No, no. In my office let's say in -- my
13 office first was in Chambers Works, it was in New
14 Jersey, and then it was in Barley Mill Plaza. I
15 enjoyed the people that were there and they
16 appreciated having somebody there to help them.
17 At that time sulfuric acid plants started
18 experimenting with having a junior engineer
19 on-site, what we call an ATO, assistant to
20 operations; but, essentially it's a junior
21 engineer fresh out of college and that was
22 another opportunity for me to mentor him and so
23 that eventually he would be the day-to-day
24 engineer to assist the acid plants.



1 The acid plants traditionally had
2 been you burn sulfur to make SO₂, to make SO₃, to
3 make sulfuric acid, they're fairly simple. Red
4 Lion is a spent acid regeneration plant. We
5 sometimes call it SAR. So, there is a little
6 more complexity to it and it is valuable to have
7 technical resource on-site.

8 Q. During that time was Elizabeth Cromwell
9 working at Red Lion do you recall?

10 A. She worked towards I think the end of
11 2009. During that time that was where the
12 refinery was having a lot of problems. They were
13 having -- when we get feed from them, so, it was
14 apparently the end of 2009 that either the
15 refinery production would be curtailed severely
16 and they there were already rumors for sale. So,
17 Elizabeth came towards the end of the Red Lion
18 operation before it was idle.

19 And, again, I don't remember exactly
20 the dates, but I think when she came within
21 months they announced the closure of the refinery
22 and we were scrambling trying to figure out
23 whether Red Lion can put utilities in place, how
24 much it would cost, what it would take to make



1 Red Lion stand alone, which was not
2 cost-effective, and then scrambling trying to
3 mothball the plant so that it would be preserved
4 given number of years so that we can restart it.

5 Q. Has that happened yet, the restarting of
6 Red Lion?

7 A. Yes, it was restarted in 2011. I think
8 it was a year, less than a year, yes. So, she
9 was there for the later part and then when it was
10 idle and then she was reassigned back to
11 Burnside.

12 Q. When did the plant go idle? Was it 2009?

13 A. No. It went idle somewhere the first
14 quarter of 2010. Part of the reason I look for
15 some of the files was to try to remember dates
16 because I knew it was in 2010, but, again,
17 because there was an overlap too. At that time
18 we knew we were going to idle the plant at Red
19 Lion. The ATC engineer that was supporting
20 Burnside for a number of years had retired in
21 2009 and they had an interim engineer assigned to
22 Burnside.

23 So, during all that time between
24 making preparations to idle Red Lion my



1 supervisor already talked to me about, okay, what
2 would be my next assignment, you know. So, at
3 that time I had two choices, either support
4 Burnside plant or support Morses Mill, which is
5 another acid plant we own in New Jersey.

6 Q. The New Jersey plant is called Morses --

7 A. Morses, M-I-L-L, M-O-R-S-E-S.

8 Q. Who was your boss that gave you that
9 choice?

10 A. Kelli Kober, K-E-L-L-I, K-O-B, as in boy,
11 E-R.

12 Q. And, so, I take it you chose to support
13 Burnside?

14 A. Yes, primarily because Morses Mill was
15 very similar in configuration as Red Lion and the
16 truth of the matter is that I probably should
17 have left Red Lion after a year because I already
18 knew enough of the plant, but I liked the people
19 so much that I --

20 Q. You stuck around?

21 A. I stuck around. So, after they idled the
22 plant and the plant manager reassigned to another
23 plant essentially there were only four or five
24 people, four or five operators, that I had known



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1 since the plant was started up that were left
2 guarding the plant, waiting for it to restart.

3 So, Morses Mill was very similar to
4 the Red Lion configuration, but Burnside had a
5 totally different configuration, which make it
6 attractive for me to learn something new. Also,
7 they had the technical need, they needed some
8 help because the plant in 2009 -- this is
9 Burnside -- in 2009 was transformed from what we
10 call a single absorption plant into a dual
11 absorption plant. Red Lion is a dual absorption
12 plant. So, I was able to use part of my
13 experience at Red Lion and apply it to Burnside.
14 That's the type of discussion that I had with my
15 supervisor that prompted me to choose Burnside.

16 Q. Did Red Lion have some of the same type
17 of equipment that Burnside ended up putting in
18 when it went to the dual absorption?

19 A. Yes, some of the same type of equipment.
20 Essentially when you go from a single absorption
21 to a dual absorption you add another absorption
22 tower. So, there are -- and you add, you know, a
23 couple more gas to gas heat exchangers.

24 Q. So, at Burnside when it went to a double



1 absorption design, that's when it added the HIP
2 and the CIP?

3 A. Yes.

4 Q. But the converter had been there a while,
5 is that right?

6 A. Right, the converter remained the same,
7 but the way they're connected, what we call
8 configured, is changed.

9 Q. When you were at Red Lion, did you
10 encounter some problems with the CIP and the HIP
11 type of equipment they had?

12 A. With the CIP, but I need to clarify that
13 even though the nomenclature calling it --
14 essentially it's a gas to gas heat exchanger, we
15 are trying to recover heat from the hot gas and
16 impart it to the cold gas. So, the function is
17 the same, but the design is different. At Red
18 Lion the CIP is what we call a shell and tube
19 heat exchanger and it is made of carbon steel
20 construction, which is the more conventional type
21 of design for that kind of service.

22 The CIP at Burnside is a Monplx
23 design, M-O-N-P-L-X. It's registered or
24 trademarked, we should know, we own them now, but



1 it's stainless steel plate heat exchanger. They
2 both function, service, as a gas to gas heat
3 exchanger, but the design and material
4 construction are different.

5 Q. Is that just the CIP or the HIP and the
6 CIP that have those different --

7 A. The HIP on both plants are same type,
8 they're Monplx. Red Lion when we compare the
9 CIP, Red Lion is shell and tube and Burnside is
10 Monplx, but both plants have Monplx type heat
11 exchangers for servicing as a HIP.

12 Q. So, did you encounter any problems at Red
13 Lion with the CIP there?

14 A. Yes. Coal -- sorry, sometimes I -- CIP
15 is an acronym that is primarily used at Burnside.

16 Q. Oh, okay. It's a cold interpass
17 exchanger?

18 A. I call it cold IP heat exchanger because
19 that's when I started at Red Lion. Sometimes
20 I'll use both interchangeably, but the CIP at Red
21 Lion or any kind of cold IP heat exchanger is
22 susceptible to corrosion. It's just the nature
23 of the process. So, for the case at Red Lion we
24 had -- we had, let's say, two type of problems.



1 Usually there is fouling because sulfates form
2 internally and that tends to increase the
3 pressure drop in an IP heat exchanger and
4 ultimately sometimes leads to internal or
5 external corrosion and leaks.

6 In the case of Red Lion we had
7 internal leaks and when that happens there is a
8 bypass of gas around the converter so that it
9 leads to higher emissions, SO₂ emissions, out of
10 the stack. So, I recall probably in -- sorry, I
11 don't know exact date, but it may be closer to
12 2008 where we were noticing that stack emissions
13 at Red Lion were progressing a little higher than
14 we had normally encountered. We were still in
15 compliance, but it was just, you know, unusual.

16 Q. Going in the wrong direction with --

17 A. Well, not only going the wrong direction,
18 but part of my role was to look at the overall
19 process of the plant and I monitor the plant and
20 the plant goes through various type of feedstock
21 distributions, production rates. So, I look for
22 correlation that said, okay, if the plant is
23 operating this way this is what I expect the
24 converter will perform, which is a direct



1 relationship to the SO2 emissions and I was
2 noting something unusual.

3 So, we did a number of tests and I
4 came to the conclusion that there was an internal
5 leak. You know, again, there are tubes inside a
6 shell, so, there was leaking internally very
7 slowly. The next scheduled shutdown, which was a
8 couple months later, I convinced the operations
9 manager to schedule an internal inspection and
10 sure enough we found about ten tubes that were
11 leaking and there are probably several hundred
12 tubes. So, we plugged those, restarted, and the
13 nice thing about it is we did not wait until we
14 had emissions problems before taking unscheduled
15 shutdown to figure out the problem.

16 You know, I felt good about it that
17 we out of looking at trends we identified
18 potential problem , run some tests to at least --
19 if not confirm, at least give a high probability
20 that there was a leak and, surprisingly enough,
21 convincing the operations manager to schedule a
22 leak check because that takes resources. In a
23 turnaround you only have limited time to do a lot
24 of work. We try to do the complete turnaround



1 usually in two weeks, but I felt good that he
2 trust me enough to spend the resources and it
3 paid off.

4 Q. So, he was happy too, I guess, to get it
5 addressed early?

6 A. I think we were all happy. Within DuPont
7 and the old assignment that I had it was probably
8 one of my best assignments. That's why I stuck
9 around longer than I needed or had to.

10 Q. You were mentioning two types of problems
11 that any cold IP exchanger can have and one was
12 the fouling because of the sulfates occurring
13 internally, which can cause internal leaks or
14 external leaks.

15 A. Right, right.

16 Q. And then what is the second type of
17 problem?

18 A. Oh, well, the fouling is a problem
19 because it's kind of restricts the gas flow
20 because the pressure drops which makes the
21 compressor in the plant work harder, which
22 requires more -- so, that's the first problem,
23 which leads to the second problem because the
24 sulfates eventually become corrosive and that



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1 leads to corrosion and the leaks. That's what I
2 meant.

3 And because the cold IP heat
4 exchanger, the CIP, handles a more extreme
5 temperature of hot gas -- well, it doesn't handle
6 the hottest gas, but the difference in
7 temperature between the hot and the cold gas is
8 the highest. That tends to stress the equipment.
9 So, usually it's a normal problem in acid plants
10 to have leaks and having to address them on a
11 regular basis.

12 Q. And even though the CIP, the cold
13 interpass exchanger, at Burnside was made of
14 stainless steel did it still have the same
15 concerns or issues?

16 A. Yes. In acid plants or any place
17 internal you cannot stop corrosion. There is
18 nothing that will stop corrosion. There are, but
19 it's too expensive. Essentially you try to
20 control, we try to control the corrosion, try to
21 depict the corrosion rate so that we can forecast
22 when we need to replace equipment.

23 Q. All right. So, you eventually get
24 assigned as the engineer for Burnside --



1 A. Mm-hmm.

2 Q. -- which that must have started around
3 the end of the first quarter 2010, does that
4 sound right?

5 A. Yes.

6 THE WITNESS: Can I stop a minute?
7 My wife sent me a text message so I just want to
8 make sure.

9 (Brief recess.)

10 EXAMINATION

11 BY MS. BARNEY:

12 Q. When you switched over to being assigned
13 to Burnside, did you sort of envision the same
14 role that you had had at Red Lion that you would
15 have at Burnside?

16 A. You mean roles and responsibilities or
17 same working environment?

18 Q. Roles and responsibilities.

19 A. Yes, the roles and responsibilities I
20 expected it to be the same. Again, and I should
21 say, that the roles and responsibilities are not
22 written in stone.

23 Q. Right.

24 A. Okay? I found out -- I mean, at DuPont,



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1 I don't know how many groups, and the only time
2 the roles and responsibilities comes into play as
3 a written document is when you interview for the
4 job to explain to the new candidate or when they
5 do performance reviews or whatever. The way my
6 experience of roles and responsibilities in
7 assignments within DuPont or the companies is
8 between, let's say, myself and the person I work
9 for. Sometimes not even my supervisor.

10 For example, at Red Lion I had asked
11 -- my supervisor I had very little interaction
12 with, okay? So, I look at the plant manager as
13 to giving me direction as to how I want to help
14 them. And he told me, he said, You are doing
15 fine, just keep doing whatever you think is
16 needed and I'll tell you when I need you to do
17 something else. I had a lot of freedom at Red
18 Lion to decide what I needed to do and how I
19 could help the plant.

20 Q. So, the roles and responsibilities, I
21 take it, sort of were affected by the working
22 relationship?

23 A. It is. I mean, I was on very good terms
24 with the plant management. So, I kid them, I



1 said they sometimes treated me as a junior
2 engineer. They asked me stuff that they should
3 be asking the ATO engineer, but it was in good --
4 I said, Whatever you need to ask. I even
5 shoveled snow at the plant one time when we have
6 I think 30 inches of snow and just kind of stress
7 out the plant. Acid plants don't like cold
8 weather.

9 So, my roles and responsibilities the
10 way I view it, the way that have worked well for
11 me, is based on a mutual understanding, not
12 anything written that I needed to do. So, I
13 looked at the roles -- a primary goal for my
14 support of the acid plant is how can I help them
15 to achieve production goals, operate within their
16 constraints, you know, limitations with
17 equipment, try to address business needs and for
18 the most part Red Lion through the two plant
19 management changes involved me very much in those
20 type of tasks, even things that are not, let's
21 say, technically my responsibility.

22 For example, Red Lion had some --
23 initially some air permit problems because we as
24 a company underestimate the emissions and I was



1 able to help identify what the errors were in our
2 original calculations. There was nothing wrong
3 with the design of equipment, it wasn't purposely
4 trying to increase emissions because we wanted to
5 increase production. I was able to provide on a
6 technical basis where we made a mistake in the
7 calculations and provide a process basis how we
8 can limit the emissions while trying to maintain
9 close to production rates that the plant was
10 designed for.

11 Q. At Red Lion the plant management really
12 sort of embraced you as part of the team to solve
13 a problem?

14 A. Right, right.

15 Q. That was not really the case as much at
16 Burnside, is that right?

17 A. No. Again, maybe because Red Lion was
18 highly unusual. I mean, even the plant manager
19 at Red Lion at the time was John Jeffries and he
20 was plant manager at another DuPont acid plant.
21 So, initially he was surprised how much time I
22 would spend at the plant and how much I
23 interacted with them. So, I look at it and the
24 thing you mentioned earlier it's a mutual



1 understanding, respect and appreciation. You
2 know, I liked the work environment. They found
3 that I can be useful. It worked out pretty well.

4 So, I mean, even, again, John
5 Jeffries told me the interaction that the ATC
6 engineer with the plant was highly unusual from
7 his experience with the ATC engineer that was
8 assigned to his previous plant. So, I didn't
9 really expect to be embraced at Burnside the way
10 I was at Red Lion, but, again, the working
11 relationship was totally different.

12 Q. And, so, how did that impact the
13 day-to-day doing of your job?

14 A. Well, the day-to-day didn't really impact
15 it too much. I looked for things that I thought
16 needed to be done. I had a good working
17 relationship with Percy Bell. I met Percy
18 probably in 2006 or '07 when I visited Burnside.
19 At that time there was another ATC engineer and I
20 went there to give him a hand and I met Percy.
21 So, immediately Percy and I became friends. We
22 are still friends. We talk on a regular basis
23 weekly and, you know, although he has not met my
24 family, I met his family.



1 So, working relationship, I mean, the
2 day-to-day stuff and I tried to go there maybe
3 one week out of the month, you know, just trying
4 to get everything done within five days so that I
5 limit my travel because I was in Wilmington. So,
6 I just pick and choose based on what I saw the
7 plant needed because I do have access or I did
8 have access to the DCS historian.

9 Q. And that's on the computer?

10 A. Right. I can see operating parameters
11 that are recorded and I can also see the
12 operators' control screens. So, I can generally
13 -- you know, even though it doesn't give a
14 complete picture of how the plant is operating it
15 gives me a very good picture of how the plant
16 operates, look at trends, production rates,
17 emission rates, utilities consumption. So, I
18 pretty much was pick and choose whatever was
19 needed.

20 During my tenure at Burnside there
21 was a turn round, so, I did spend the whole month
22 except for two days, I think. I spent the whole
23 month of April 2011 I think, yeah. So, with
24 day-to-day I usually interacted with Percy and



1 the operators.

2 Q. And Percy was the day operator, right?

3 A. Well, I don't remember what his exact
4 title is. I look at him as a lead operator. He
5 has -- not only he has the most experience in
6 operations of the plant I think he has -- he
7 fully understands the plant. I had told Percy
8 one time that under different circumstances he
9 would have been a very good engineer. In some
10 respects he is a better engineer than some
11 engineers I know of, but he just lacks -- he
12 didn't have a piece of paper. I never ask him
13 why he didn't pursue an engineering degree.

14 But in terms of who I rely on at
15 Burnside to give me information, again, the DCS,
16 what I can see in the computer in the trends,
17 it's like just a snapshot. You can't not fully
18 understand what is going on at plant, what is
19 causing it, without talking to somebody who is at
20 the plant.

21 Q. You didn't have any problems with Percy
22 embracing your assistance and working with you?

23 A. Yeah, I think we embraced too much. We
24 would go out to lunch and I would take him out to



1 dinner and I met his family. I'm sure you have
2 met him. He is a very personable guy.

3 Q. Yes.

4 A. He is a fine person. Most of the
5 discussions we have now that I don't support
6 Burnside relate to his family, my family. I told
7 Monique that I called him a couple days ago to
8 check on him because he had like a stomach virus
9 and it was unusual for him, he took sick days I
10 think two or three days in a row.

11 Q. And he has been there since -- 30
12 something years I guess?

13 A. I'm not sure if it's 34 or 33. It's more
14 than 30 because there were other operators that
15 had 30-some years that have since left. But,
16 again, a lot of operators know how to operate the
17 plant, they know how to push buttons, turns
18 valves and all that, but Percy has a very good
19 understanding of the process. So, he can
20 anticipate if he makes a change --

21 Q. What is going to happen?

22 A. -- what are the consequences as opposed
23 to waiting for the consequences to happen. I
24 would like to say that he actually makes my job



1 enjoyable and that's good.

2 Q. That is good. So, Percy, you would say
3 is the most knowledgeable in the process, but is
4 he the person who can make the decisions about
5 when to shut down or how to schedule an
6 inspection?

7 A. No. I'll say technically anybody at the
8 plant, technically, has the responsibility or
9 power to shut down the plant if he or she thinks
10 it's warranted.

11 Q. And that's the sort of emergency
12 shutdown, push a button, basically?

13 A. Well, that's one way to shut down the
14 plant, okay? But what I'm saying is at DuPont we
15 like to say that if you think -- if you don't
16 think something is safe, we are told that you
17 have the authority to stop the job and I include
18 the plant, to stop the job, if you don't think
19 it's right, whether it's safety, even ethical,
20 anything that you -- you personally don't feel
21 safe you can shut it down. Now, whether or not
22 that's implemented in practice is based on the
23 individual, right?

24 But, as far as what authority Percy



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1 has or responsibilities I still -- even after the
2 two plus years I was there I'm not sure what his
3 responsibilities are because sometimes he takes
4 responsibilities that are not -- I think they go
5 to him because people know he knows. So, he
6 takes on himself more than, let's say, his roles
7 and responsibilities that are called for. I
8 don't know if that answers your question.

9 Q. Yes. And if he thought that there should
10 be a shutdown, let's say, to address leaks on the
11 CIP, the HIP or the converter he would need the
12 support of his plant manager to actually make
13 that happen?

14 A. Right, unless -- to me an emergency, for
15 example, although it has never happened, you
16 know, if a duct ruptures or if an acid pipe
17 breaks, that is a real emergency that, you know,
18 operators should feel the need to shut down the
19 plant immediately. They don't have to check for
20 it. If it's not imminent danger, say you are
21 addressing if you see a leak, a gas leak, it
22 depends on the magnitude of the leak, right? If
23 the leak is a result of a duct rupture, which is
24 tremendous, but, if a leak due to corrosion, it's



1 incumbent on Percy, the operators, to alert
2 whether it's the operations manager or the
3 maintenance manager that there is a problem.

4 And it's incumbent upon the
5 maintenance supervisor if they're not on-site to
6 either try to ascertain them on the phone or go
7 to the plant to determine whether it needs to be
8 shut down or not, but not every leak that an
9 operator or Percy in particular would see, again,
10 would they go and shut down the plant. To me
11 even emergency shutdown, pushing a button to shut
12 down the plant, that also requires a criteria.
13 If you need to shut down the plant, you see if
14 you can do a controlled shutdown as opposed to
15 pressing a button, which would just shut the
16 equipment and trap the gases.

17 An analogy in emergency you try to
18 apply car brakes to stop the car, but, if the car
19 brakes don't work, then you try to use the
20 parking brake; but, you don't use the parking
21 brake to try to stop the car every time just
22 because it's convenient.

23 Q. Right. I was asking, I guess, more along
24 the lines if Percy felt like there were leaks in



1 the equipment, in the CIP, the HIP, the
2 converter, that required repairs during a cold
3 shutdown and he would need the plant manager or
4 others above him to support the decision to go
5 into a cold shutdown to do the repairs?

6 A. Yeah, actually, the role I see for the
7 plant operators, again, not just Percy in
8 particular, is if there are leaks, if there are
9 problems with equipment, is to alert their
10 supervision. Again, that would be operations,
11 not even the plant manager directly. The chain
12 of command is the immediate supervisor. For
13 Percy it would be Elizabeth. For the maintenance
14 folks it would be Gene Clemons.

15 It's up to them to decide the, you
16 know, not only by themselves, they can draw
17 resources. You know, within ATC we have also
18 consultants, internal consultants, that
19 specializes in materials construction,
20 refractory, with rotating equipment that the
21 plant can draw upon and say we are having this
22 problem; or, even if it's an obvious leak, we
23 have experts within the company say this is how
24 you can repair it.



1 So, again, the role of the operators
2 and maintenance folks they cannot -- even myself
3 I cannot -- even if I see the leak I don't know
4 the best way to repair it. I know it needs to be
5 repaired, but that's why it's incumbent upon
6 management to draw the proper resources based on
7 their experience, based on operating practice for
8 an acid plant, who to bring and the procedures.
9 Some of them we have standard procedures how to
10 repair.

11 MS. BARNEY: I think it's the ETA
12 time for your wife.

13 THE WITNESS: Thank you.

14 (Brief recess.)

15 THE WITNESS: Normally I will have
16 like spread sheets for calculations and this just
17 happened to be the last type of calculation I had
18 done for gas leaks. I mean, I don't know the
19 calculations for Red Lion and some of the
20 calculations don't pertain even specifically to
21 actual leaks. We run a hazards analysis where we
22 try to predict based on different conditions what
23 the potential hazards would be. So, this
24 happened to be the spread sheet I was using and I



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1 forget the date, but.

2 MS. WEINER: Here.

3 THE WITNESS: At that time Kerry
4 Long, who was the --

5 MS. BARNEY: Can we wait? I'm sorry,
6 I need to say something for record real quick.
7 During the break, Mr. Chu, you were able to
8 obtain copies of documents that you looked at
9 recently before your deposition?

10 THE WITNESS: Right.

11 MS. BARNEY: So, I'll just go ahead
12 and mark those as Exhibit 1 and 2. I will mark
13 the one-page document as Exhibit 1 and the multi-
14 page document that has preliminary conclusions on
15 it as Exhibit 2.

16 (Chu Deposition Exhibit Numbers 1 and
17 2, respectively, were marked for identification.)

18 MS. BARNEY: If it's okay, I'll come
19 back to these in a little while for some
20 questions for you and I'll try to pick up where
21 we left off. Is that all right?

22 THE WITNESS: Yes.

23

24



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1 EXAMINATION

2 BY MS. BARNEY:

3 Q. We were talking earlier about the
4 Burnside plant and sort of your interaction with
5 Percy at the plant and his role. It's your
6 understanding I guess because you have such a
7 close relationship with Mr. Bell that there have
8 been leaks on the CIP, the HIP and the converter
9 for almost two years. Is that your
10 understanding?

11 A. Yes.

12 Q. And those leaks have been pretty
13 continuous with patch jobs here and there, but
14 never really being able to shut down the leaks,
15 is that right?

16 A. Yes. They were having problems trying to
17 locate them and even if they locate them to try
18 to be able to make a lasting repair. They tried
19 to control it with hoses to suck the leaks into
20 the drying tower.

21 Q. What is your understanding of the hose
22 apparatus that they have attached to the
23 equipment?

24 A. What do you mean?

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1 Q. The black plastic hoses I think is what
2 they are.

3 A. Mm-hmm.

4 Q. Are you familiar with the hose system at
5 Burnside?

6 A. Yes.

7 Q. Did you come up with that system?

8 A. No.

9 Q. Have you suggested the use of that system
10 at any other plant?

11 A. No.

12 Q. Did you suggest that system at Burnside?

13 A. No.

14 Q. So, that was a system that was sort of
15 created I guess by Gene Clemons, the maintenance
16 supervisor?

17 A. No, it was before that. I don't really
18 remember when it was put in place. I mean, the
19 concept is all acid plants at one time or another
20 would have a gas leak and even at Red Lion we had
21 an external leak on a hot heat exchanger, you
22 know, and the only way to control it while we
23 plan permanent repairs is temporarily run a hose
24 that connected the drying tower, which operates



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1 under vacuum conditions, temporarily connected to
2 the location of the leak.

3 In fact, the person who suggested
4 that at Red Lion originally was maintenance
5 supervisor at Burnside.

6 Q. Oh, okay. Who was that?

7 A. James Harman, H-A-R-M-A-N. So, it is an
8 effective way to temporarily control a leak,
9 particularly a leak that is not very large, but
10 it is not intended to be a permanent solution.

11 Q. When you say temporarily at Red Lion, how
12 long did they use that hose system at Red Lion?

13 A. I'm talking about weeks, less than a
14 month. I don't remember exactly. What it
15 entailed is that I think the leak happened in the
16 wintertime. Sulfuric acid, strong sulfuric acid,
17 has a tendency to freeze and that causes a lot
18 more problems. So, we had to pick a particular
19 time of the month when the weather was warm
20 enough and we had the necessary resources, people
21 and procedure, to address the leak.

22 Q. And at Red Lion it took almost a month to
23 do that, to get that right time?

24 A. I'd say less than a month, but I don't



1 remember exact dates, but, we run the hoses,
2 again, it was temporarily. There is a manifold
3 system that connects certain process equipment,
4 specifically pump tanks, to control vapors that
5 are given off by the acid that is contained in a
6 pump tank. Those are manifolded into the drying
7 tower. That's a normal design.

8 Q. That was at Red Lion?

9 A. Normal design in any acid plant that has
10 a pump tank. You look at the pump tank as a
11 reservoir that pumps, draws acid from and
12 circulates around the towers; but, of course, if
13 you have strong acid, you have tendencies of off
14 gases like SO₂ to be given off. So, typical acid
15 plants control that by installing ducts that
16 connects the pump tanks into the drying tower.
17 So, at Red Lion what they did temporarily is out
18 of the same manifold connect hoses to the source
19 of the leak at the gas to gas heat exchanger, hot
20 heat exchanger. So, it was put in place to
21 control the leak, plan and execute the repair,
22 and taken off.

23 Q. And I take it you were involved with the
24 plant when that occurred at Red Lion?



1 A. Right, I was responsible for the plant at
2 that time.

3 Q. About what year was that or how far
4 before they started going into idle would you say
5 that was?

6 A. Oh, that would have been I'll say if not
7 2008, 2009. I will say about a year before we
8 idled it. I'll say 2009.

9 Q. So, you were okay with using that system
10 as a temporary measure?

11 A. Mm-hmm.

12 Q. Was there any discussion when you all
13 implemented that at Red Lion as to how long, you
14 know, what was the outside time you would want to
15 operate like that?

16 A. Not a specific time, but it was the plant
17 manager's directive that this was temporary, that
18 we -- you know, we could not operate like that.
19 So, eventually we had to take a cold shutdown
20 because we did try to make some repairs, but it
21 continued to leak. We found out that there was a
22 problem in the initial installation of the heat
23 exchanger. They should have welded inside the
24 duct as well as outside. They only welded



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1 outside. So, we had to take a cold shutdown,
2 figure out a way to weld it internally and did
3 the repair, stopped the leak, discontinued the
4 hoses.

5 Q. So, at Red Lion they took a cold shutdown
6 for the specific purpose of addressing that leak?

7 A. Right.

8 Q. How long were they shut down?

9 A. I don't remember exactly, but I would say
10 a week.

11 Q. Is that the only time that you can recall
12 Red Lion having a cold shutdown to repair a gas
13 leak?

14 A. Yeah, that's the only one that I recall.
15 Sometimes we do have leaks in the duct, but
16 because part of that plant is in a vacuum we are
17 actually leaking air into the system. So, in
18 those cases sometimes we can afford to wait a
19 little longer.

20 Q. That's where you are sucking the
21 atmosphere into the system --

22 A. Right.

23 Q. -- rather than putting gas out?

24 A. Right, right. Again, I'm not really



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1 involved with the maintenance side of the
2 repairs. There might be other leaks that can be
3 addressed by just if not repairing the leak
4 specifically, you can build boxes around it to
5 contain it for a period of time until the next
6 scheduled shutdown where you remove the
7 containment box and effect a more permanent
8 repair of the duct or the heat exchanger, but,
9 that's the only time I recall using the hoses,
10 again, for the temporary repair.

11 Q. That you talked about earlier?

12 A. Right.

13 Q. During that time period, that three to
14 four weeks at Red Lion or so, when the hose
15 system was used were those black plastic
16 corrugated hosing material?

17 A. Yeah, we started with that and sometime
18 after that we had to switch to stainless steel.
19 The reason being that if you use the black hoses
20 to stop leaks if you are sucking things like S03,
21 S03 will react with moisture in the atmosphere
22 because you don't selectively just suck a leak.
23 You suck the neighboring area. So, you end up
24 sucking a portion of ambient air that contains



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1 moisture and moisture water in S03 form, sulfuric
2 acid, and reaction of S03 and water generates
3 heat and those corrugated pipes are not designed
4 to handle the heat. So, eventually you can start
5 melting them or putting a hole in it.

6 That's why we never think of it as a
7 permanent solution. It's just a way to control
8 it where you can better plan a repair.

9 Q. So, when you all switched to the
10 stainless steel hose system at Red Lion during
11 that temporary time period, was all the hose
12 switched to stainless steel?

13 A. No. Only the areas where they're most
14 susceptible to the high temperature, but even
15 stainless steel will be susceptible to corrosion
16 at high temperature in acidic conditions.

17 Q. During that time period at Red Lion, that
18 less than a month or so, did you have to run any
19 calculations about the amount of gas that was
20 escaping during the leak?

21 A. No. Because for the most part the leak
22 was not big enough and I guess since the gas duct
23 that was leaking contained some S03 you can
24 visually see if it was leaking even if it wasn't



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1 contained. Of course, there would be times
2 because the wind would move the hose or something
3 it was part of the procedure was for the
4 operators to inspect the area on a regular basis
5 and position the hose properly to make sure that
6 that wasn't the case. The leak that I had in
7 mind is it was an area that was at least
8 accessible and visible to the operators for them
9 to monitor and try to control.

10 Q. So, at Red Lion the operators themselves
11 actually were allowed and expected to adjust the
12 hoses?

13 A. Yes, especially at night in most acid
14 plants we only have two people, two operators, no
15 maintenance folks at night. So, they are the
16 only ones on-site that will do it. The roles of
17 operators, maintenance, varies from plant to
18 plant what each is supposed to be doing. Laying
19 out the pipe or the temporary duct, that would be
20 the role of the maintenance guy. Monitoring and
21 making adjustments where there are no maintenance
22 folks you have to rely on the operators to do
23 that.

24 Q. That's how they did that at Red Lion?



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1 A. Right.

2 Q. I guess the leaks of S03 are more visible
3 at night, is that right?

4 A. No. Leaks for S03 are visible more so in
5 the daytime because when S03 reacts with moisture
6 in air it generates like a mist. What you are
7 seeing is an acid mist. You are not seeing
8 actually the S03. You are seeing the product of
9 the S03 reacting with the moisture. S02 are
10 gases that you cannot -- S02 you cannot visually
11 see. We have monitors that detect S02 gas
12 concentrations especially at ground level.

13 Q. A lot of folks have talked about it being
14 easier to see S03 at night, but that's not your
15 experience, you think it's easier to see -- I
16 think it was the contrast with maybe the dark and
17 white fumes that they thought it was easier to
18 see.

19 A. Well, at night it may be visible if you
20 have light, not just because it's dark. In the
21 daytime you may on a cloudy day if the leak is at
22 a high level and you are looking against a cloudy
23 sky, you may not be able to see; but, again, it
24 depends on I will say the leakage rate, but I



1 can't attest to the fact that you can see it
2 better at night.

3 Q. I guess a light shining on it at night
4 might reflect off of the vapors, is that right?

5 A. Yeah, right, right.

6 Q. It's more the light that you have at
7 night than it is the night darkness?

8 A. Right. The darkness wouldn't help. If
9 it's completely dark, no, I can't see -- you have
10 to visually be able to shine and reflect against
11 it, the vapor.

12 Q. Do you recall at Red Lion during that
13 temporary time gas clouds of the S03 traveling
14 across the plant site, like from one part of the
15 site to another?

16 A. No. That would be very bad. Again,
17 because, again, ground level, that type of
18 emissions, particularly at Red Lion will affect
19 the operators and maintenance folks first. Yes,
20 we had the refinery right next to it on the fence
21 line, but there was a buffer of tanks and other
22 equipment before it gets to the refinery
23 personnel and there is not a lot of surrounding
24 area -- there are not a lot of people or



1 residents around the surrounding area.

2 Q. So, I guess at Red Lion you never had to
3 or they never had to kind of clear workers away
4 from one area of the plant because a gas cloud
5 was moving that way from these leaks?

6 A. Not from leaks, from leaks from ducts if
7 you are referring to that. They had to clear,
8 for example -- yeah, there would be occasion
9 where they have to clear personnel because, let's
10 say, the spent furnace is used to decompose
11 sulfuric acid. We burn fuel gas to create an
12 environment where the spent sulfuric acid
13 decomposes back to SO₂, some SO₃, water, oxygen,
14 we burn the carbons in the spent acid.

15 So, in situations where, for example,
16 there is an electrical power failure and it shuts
17 down the plant, the furnace under normal
18 operation is operating under a slight vacuum so
19 the gases are drawn to the rest of the plant;
20 but, when the process shuts down the blower
21 starts coasting down and loses power, then you
22 run into a situation that the furnace becomes
23 positive. So, the hot gases start expanding out.

24 So, in that type of situation, yes,



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1 there have been quite a few electrical power
2 failures in the Delaware City area that causes
3 the operators to get out of that area because you
4 can't seal the furnace completely. There is
5 going to be some gases escaping out of it.

6 Q. At Red Lion I guess during this temporary
7 period where they had the hoses on some of the
8 leaks on the equipment at Red Lion I take it the
9 employees didn't have to dodge the gas --

10 A. No.

11 Q. -- and kind of watch the wind socks to
12 figure out what direction --

13 A. Well, you always watch the wind sock. I
14 mean, it's one of the first things we try to
15 teach people, always know the wind direction,
16 because, you know, unforeseeable situations
17 happen; but, no, they did not have to dodge, no.
18 You try to avoid the area if you don't need to be
19 there, all right, but on the same token you have
20 to have people go by there and inspect to make
21 sure it's -- the hose was in the proper place and
22 it was contained.

23 The gases that were leaking that
24 would be a combination of S02 and S03. S03 you



1 see visually and S02 we had monitors on the fence
2 line. That was an agreement that we had with the
3 refinery, again, in case of emissions problem
4 that it would be detected at the fence line
5 before it gets to personnel in the refinery. I
6 would not want to be in an area where I have to
7 keep dodging clouds of S02 or S03 gas.

8 Q. At Red Lion during that three to
9 four-week period did any employee complain that
10 they had walked into a gas cloud of S03 or S02?

11 A. Not that I recall and, again, it would
12 not be my -- I would not be the person that they
13 come to.

14 Q. But you never heard that the leak got so
15 bad that somebody claimed they had walked into
16 it?

17 A. Again, not that I know of, no.

18 Q. Do you know how many feet of hose they
19 used for that temporary fix in Red Lion about?

20 A. If I had to guess, less than 50.

21 Q. Did they have to change out the hose
22 material from time to time? Did it actually melt
23 down and have to be replaced?

24 A. Yes, yes. Again, that was the purpose of



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1 the inspection.

2 MS. BARNEY: Let's take a lunch
3 break.

4 (Lunch recess.)

5 EXAMINATION

6 BY MS. BARNEY:

7 Q. So, you started being involved with
8 DuPont Burnside when? Did we say middle of 2010?

9 A. Somewhere in the first quarter of 2010.
10 Again, it was a transition where I felt I need to
11 finish some stuff at Red Lion and get the plant
12 in idle and then switch over.

13 Q. When you first started working with
14 Burnside these leaks we were talking about, which
15 started two years ago, those were not happening
16 then?

17 A. No, no. Again, the plant at that time
18 was probably less -- the CIP and HIP were less
19 than six-months old at that time.

20 Q. Were you worried at that point about how
21 the CIP and the HIP were going to perform because
22 of your experience at Red Lion?

23 A. Worried not in the sense that I expected
24 the CIP to have some leaks, but worried in the



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1 sense that the Burnside was a retrofitted plant,
2 was converted from single to dual absorption,
3 that the distance between the IPAT, interpass
4 absorption tower, and the CIP is unusually long.
5 They usually are a lot closer together. So, the
6 chances for the gas to cool down and the acid
7 mist that would be in the gas coming from the
8 absorption tower to condense was more likely. In
9 fact, we made some changes to alleviate that.

10 Q. What changes?

11 A. I had the plant install what they call a
12 boot, B-O-O-T, which is just a small vessel to
13 help catch the acid that might collect in the
14 duct before it gets to the CIP.

15 Q. I would like to show you a document that
16 is Bates labeled DSF618 through 634 and this is
17 an e-mail on top from Kirk Bailey to you and Dan
18 Monhollen and some other folks --

19 A. Mm-hmm.

20 Q. -- dated January 26, 2012. Do you
21 remember seeing this -- and attached, I'm sorry,
22 attached is the December 6, 2011 DuPont Burnside
23 Pegasys test.

24 A. Yeah. I'm sure I'm seen it, yes.



1 Q. So, that Pegasys report looks kind of
2 familiar?

3 A. Yes.

4 Q. It's ringing a bell?

5 A. Well, yes, yes.

6 Q. That report states that it was run based
7 on data from December 8, 2011, I believe, if you
8 look at page 625 --

9 A. (Witness complies.)

10 Q. -- in that first paragraph.

11 A. Mm-hmm, yup.

12 Q. Is that your recollection?

13 A. Yes.

14 Q. Let me show you an e-mail --

15 MS. BARNEY: I guess we should mark
16 this first document as Exhibit 3.

17 (Chu Deposition Exhibit No. 3 was
18 marked for identification.)

19 EXAMINATION

20 BY MS. BARNEY:

21 Q. And then I'll show you an e-mail, dated
22 December 9, 2011, which seems to be referring to
23 that report. I'll show you what is Bates
24 numbered 866 through 874 and we'll mark that as



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1 Exhibit 4.

2 (Chu Deposition Exhibit No. 4 was
3 marked for identification.)

4 EXAMINATION

5 BY MS. BARNEY:

6 Q. And that's a more complete copy of the
7 e-mail and that's an e-mail from you dated
8 December 9, 2011 to Dan Monhollen with a cc to
9 Percy Bell.

10 A. Mm-hmm.

11 Q. And you are commenting on the Pegasys
12 survey?

13 A. Yes. Did somebody explain what the
14 Pegasys is about and what it covers?

15 Q. No. Why don't you do that as briefly and
16 as lay person as possible.

17 A. The Pegasys test is designed to assess
18 the performance of the catalyst and the converter
19 to see how efficient it is in converting S02 and
20 S03 and, also, the secondary objective is to
21 determine whether there is internal gas leaks in
22 gas to gas heat exchangers. The reason that's
23 important is because internal gas leaks affect
24 the conversion performance.



1 So, it's a very good tool that we use
2 to try to determine and try to ascertain which
3 areas may be causing the problems in conversion
4 and, therefore, emissions. The one thing it
5 doesn't do it cannot determine the extent or
6 magnitude of external gas leaks.

7 Q. This report would not really tell how
8 much gas was escaping from the CIP, the HIP or
9 the converter as of that date?

10 A. No, no. The reason that these e-mails
11 were exchanged and prompted is, again, I look at
12 the plant performance, conversion being one of
13 them, and the outcome, emissions, being another
14 as part of my role. The routine practice of a
15 plant is to do pressure surveys and that's what
16 you see in some of the data that I was referring
17 to thanking Dan for sending me pressure survey
18 data because 20 percent of that data has to be
19 done by field measurements. The rest I can get
20 from the DCS.

21 So, it's a normal practice that I
22 asked him to do at least once a month. Again, he
23 gives me more of a picture of how the overall
24 plant is performing, not only conversion, but



1 pressure drops, et cetera. So, my comment when I
2 say for the Pegasys that it did not indicate any
3 internal gas leaks was reassuring because as you
4 can see based on pictures that we have taken of
5 the heat exchanger this would have been during a
6 boiler leak; but, evidence that we have seen
7 during of the turnaround that we had in
8 April 2011 it was a lot of acid was being
9 collected at the inlet of the CIP and that's
10 never a good thing.

11 So, I was concerned about -- external
12 leaks we can see, you cannot ignore. Internal
13 leaks it starts slowly, but they always progress.
14 That was the purpose of the test -- I mean, of
15 the e-mail exchanges. That's the purpose of
16 running a Pegasys test.

17 Q. So, the internal leaks might have an
18 impact on the environment through the stack?

19 A. Right.

20 Q. But the external leaks have the impact
21 just at the leak point?

22 A. It still impacts the environment, but the
23 reality is we don't have monitors. You know,
24 everybody in the plant -- emissions in the plant



1 are monitored and measured based on emissions
2 points, which is the stack, so, we have gas
3 analyzers at the stack.

4 Q. And that's for SO2?

5 A. That's for SO2.

6 Q. There are no analyzers anywhere on the
7 Burnside plant or Red Lion for that matter for
8 SO3, right?

9 A. There are no such analyzers. I will say
10 there is no such on-line analyzers. Depending on
11 the plant, depending on the air permit at each
12 plant, they require on a yearly or bi-annual
13 basis to bring in third-party, you know, the
14 State being one, the plant being -- a third-party
15 contractor to do measurements that confirm that
16 the SO2 on-line analyzers are working properly,
17 as intended, and also to measure acid mist from
18 the stack.

19 A normal stack in an acid plant
20 should be clear. You should not be able to see
21 anything. There is some acid mist that comes out
22 but because the concentration is so low they're
23 not that visible. So, you have to bring an
24 analyzer that -- I don't really know how the



1 word, really I never operated one, but you
2 physically have to collect the gas -- sample the
3 gas that is coming out of the stack and you can
4 calculate the milligrams of sulfuric acid per
5 standard cubic feet of gas that is evolved out of
6 the stack.

7 Q. What kind of gas is that measuring?

8 A. Sulfuric acid mist, which is an
9 indication -- again, you cannot separate between
10 SO₃ or sulfuric acid itself, but any kind of SO₃
11 that escapes in the atmosphere, whether it's from
12 a stack or from a gas leak, eventually becomes
13 sulfuric acid because of the reaction of SO₃ with
14 water.

15 Q. So, at that point they're measuring the
16 sulfuric acid mist to try to figure out how much
17 -- they're just measuring the sulfuric acid?

18 A. There is a limit. Again, there is a
19 limit, but for most plants -- or, for all plants,
20 not only in the U. S. it's not feasible to be
21 able to measure the SO₃ acid mist continuously.
22 The procedure is complex enough and it requires a
23 lot of expertise to be able to measure
24 accurately. That's why by permit it's only done



1 once or twice a year.

2 There is a qualitative measurement
3 called opacity of the stack. If there is SO₃ or
4 acid mist coming out we call it the stack becomes
5 visible. Generally in an acid plant when opacity
6 exceeds ten percent, again, it's a visual
7 qualitative measurement. People have to go to
8 what they call a smoke school to be trained in,
9 okay, this is what it looks like, this is five
10 percent, this is ten percent, this is 20 percent.

11 So, plants are required to
12 qualitatively inspect the stack once a day, it
13 can only be done in the daytime, and see is the
14 opacity more or less than ten percent. So, if
15 it's more than ten percent, it means there is
16 some unusual amounts of SO₃ and acid mist coming
17 out.

18 Q. And the permit -- well, and I realize you
19 are not the environmental person, but you may
20 know the permits don't allow for emission of SO₃,
21 they just allow for SO₂, right?

22 A. No. The permit has emission limits. All
23 acid plants that I know of in DuPont have
24 emission limits for SO₂, acid mist, and, again,



1 opacity, again, a qualitative measurement.

2 Q. And that's coming out of the stack?

3 A. Coming out of the stack. The amount of
4 SO₂ that each plant is allowed to emit varies
5 depending on the state regulatory, depending on
6 the age of the plant; but, again, I don't know
7 right off the top of my head what the SO₃ limit
8 is, but there is an acid mist limit I understand.

9 Q. The SO₃ becomes an acid mist when it
10 comes out of the stack and that's what you are
11 saying every now and then an outside contractor
12 would come measure the mist coming out of the
13 stack?

14 A. Right, quantify it. Again, the stack,
15 once the opacity is evident we know it's acid
16 mist, but we don't know if it's because from acid
17 mist that is carried over from the tower itself
18 or the SO₃ has not been absorbed fully because
19 once it comes out of the stack you cannot
20 differentiate whether it was originally SO₃ or
21 originally sulfuric acid.

22 Q. It could be sulfuric acid itself as the
23 original chemical coming out?

24 A. Right, right, right. Either way it's not



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1 good.

2 Q. So, if you see a leak of the kind we were
3 talking about earlier that comes from the
4 equipment, that would be S03 gas because it's
5 coming out of the process at that point?

6 A. Depending on where. For example, the gas
7 that is first going to the converter only has
8 S02, but once -- and the converter, let's say, at
9 Burnside have four compartments. So, the first
10 time the gas goes into the first compartment
11 there is no S03, but after that S02 gets
12 converted to S03 and the S03 concentration goes
13 up, the S02 concentration goes down. Then, after
14 it comes out of the third pass probably
15 93 percent, 95 percent of the S02 has been
16 converted to S03 and that goes into the IPAT,
17 interpass absorption tower, where the S03 is
18 absorbed so the gas coming out of that absorption
19 tower is only S02.

20 That goes into the fourth pass where
21 the rest of the S02 is converted. So, by the
22 time it goes through everything typically
23 depending on the acid plant for a dual absorption
24 plant is 99.2, 99.7 of all the S02 has been



1 converted to S03. So, depending on where --
2 which duct, which gas to gas heat exchanger, the
3 various concentrations of distribution of S02 and
4 S03.

5 Q. If there were leaks in this equipment,
6 the HIP, the CIP or the converter, and you could
7 see them then that would tell you it's S03 gas?

8 A. Right, and S02 you cannot see. It
9 doesn't mean there is no S02, but there is out of
10 the CIP if you see S03, there is going to be S02.

11 Q. Do you recall whether around the time of
12 this Pegasys study, December 8th of 2011, whether
13 the leaks that we have been talking about
14 earlier, the ones that have lasted two years or
15 so at DuPont Burnside, had those started do you
16 know at the time this Pegasys report was done or
17 was it shortly after that?

18 A. I don't recall exactly. My guess is some
19 of them already started. That would be my guess.
20 Again, the reason for the Pegasys test was not to
21 ascertain or quantify external leaks, no. As a
22 process engineer I was more concerned with what
23 was happening internally. If I see an external
24 leak in the gas heat exchanger, I worry about



1 internal leaks too, which then it starts
2 affecting the performance of the plant and the
3 ability for the plant to operate within air
4 permit limits.

5 Q. But nobody is measuring the SO₃ gas that
6 is escaping from these internal leaks, is that
7 right?

8 A. I can't say that. That would be the role
9 of the plant. If there is a leak we can see from
10 the Excel spread sheet that I showed you earlier.
11 Those are the type of calculations that acid
12 plants will have to use to ascertain based on the
13 operating condition of the duct or equipment,
14 concentration of gases, SO₂, SO₃, pressure,
15 internal pressure, conditions, ambient
16 conditions, that's a way to estimate gas leaks.

17 Q. That's something that you said had to be
18 done at the plant level? You couldn't do that
19 from Wilmington?

20 A. No. Because the equation has a
21 theoretical basis, but the -- and things like
22 composition, temperature and pressure that can be
23 measured or estimated very accurately. Two
24 components that are critical for the gas leak



1 estimate is the size of the hole that is leaking
2 and, which unless you can visually ascertain, is
3 hard to do, and the other thing is the duration
4 of the leak, you know. Usually one can determine
5 when the leak was found you can say that was --
6 or you can determine the time where a process, a
7 procedure, was implemented to control the leak.
8 They are known as when did the leak start.

9 Sometimes you have to make an
10 estimate and that estimate can be based on, well,
11 1100 hours yesterday operator passed through the
12 area and there was no leaks. So, you can be
13 conservative and say the leak we're going to
14 start then. So, the equation, the calculation,
15 is simple, is direct, but the two unknowns -- the
16 two uncertain parameters are the size of the leak
17 and the duration.

18 That has to be determined by an
19 incident investigation where you get a group of
20 people that agree, okay, based on interviews,
21 based on what happened at the plant during that
22 period of time, what value the duration or upon
23 inspection of the gas leak is estimated as an
24 equivalent cross section of area of a dime, a



1 nickel, a dollar, whatever, so.

2 Q. If you don't have -- well, let me back
3 up. If you can't locate the leak source, the
4 crack or the hole, then you can't do a
5 calculation of how much gas is escaping?

6 A. Well, you can always do the calculation.

7 Q. Okay, you are right. In order to get an
8 accurate calculation of the gas leak you would --

9 A. I would say even if you get a size, okay,
10 you are not actually measuring the leak to say,
11 Ah, this correlation is applicable because there
12 is a lot of assumptions going into that
13 correlation; but, it's the best empirical
14 revelation that I've found and it's used
15 industry-wide to ascertain leakage from equipment
16 and ducts because measuring fugitive emissions
17 are very hard.

18 A calculation can be made, but it has
19 to be done with the realization that whatever
20 number comes out what the uncertainties are and I
21 will say a good engineer would do a calculation,
22 look at the number and say does it make sense?
23 The number he gets physically does it make sense
24 relative to the amount of gas that goes through



1 that duct equipment, does it make sense or not?

2 Q. So, are you saying that even if you know
3 what the size of the hole and the crack are and
4 you do your formula there is still some
5 uncertainties about the estimate?

6 A. Well, it's the duration. A lot of times
7 -- the other uncertainty is the duration. You
8 know when you found the leak, you know when you
9 did the containment. You contain it so that it
10 doesn't leak worse or so that it doesn't leak to
11 the atmosphere, you have contained it, for
12 example; but, the uncertain part is when it
13 started, you know, because it would be very
14 unlikely that as you pass by the leak started
15 right there and then.

16 Q. You try to go back to the last time
17 somebody saw no leak at all and --

18 A. Even that time, did he actually look at
19 that area? It's a calculation. You are just
20 trying to do the best estimate realizing what the
21 uncertainties are and I look at this type of
22 calculation, any type of calculation, does it
23 make sense, you know. It would be hard to
24 explain to somebody and say, Well, we found a



1 leak, then I'm going to assume the leak started
2 an hour ago. How do you know it's an hour ago?
3 Sounds like a nice good round number.

4 Q. Makes the math easy?

5 A. It makes the math easy because the ratio
6 in pounds per hour just multiply it out. Again,
7 I'm not trying to avoid, but in the calculations
8 as an engineer I try to see what the
9 uncertainties are. Just because I can calculate
10 the number exactly very precisely.

11 Q. If you have the information?

12 A. No, no. I can calculate very precisely
13 with a calculator, how many digits do you want, I
14 can do it to a thousandth of a gram, but it won't
15 be -- I can tell you it won't be accurate.

16 Q. So, if DuPont wanted to know how much gas
17 was escaping they would need to know the number
18 and size of the holes and cracks in the vessels,
19 right?

20 A. Yes.

21 Q. Then, you would have to run a calculation
22 for the whole duration of those leaks to get an
23 accurate estimate, right?

24 A. Yes.



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1 Q. And then you would have a complication I
2 guess of a leak that is ongoing so you might be
3 able to come up with a per hour estimate, but you
4 wouldn't have a total release estimate because
5 that is still happening, right?

6 A. I'm not sure I can --

7 Q. Well, if the leak is continuous and it's
8 ongoing, then the best you are going to come up
9 with would be a per hour rate of release, right?

10 A. Well, yeah, the calculation -- the form
11 of the calculation is a release per unit time.
12 In fact, I think it's in seconds.

13 Q. I'm sorry, okay, seconds.

14 A. Correct. It's a release per unit time.
15 So, but that release per unit time the most
16 difficult parameter to determine is the size of
17 the hole. So, if you are looking at the release
18 per unit time, you have one big unknown,
19 uncertain value. If you are looking for the
20 total amount of release, then duration is an
21 unknown. Now you have two parameters. You go --
22 I mean, to add complication to it is the size of
23 the hole never stays the same. It usually grows.

24 Q. And it expands with heat?



1 A. Expands with heat. There is a number of
2 uncertainties that, again, the solution is not
3 trying to spend a lot of time estimating it. The
4 solution is to try to do a repair. Because,
5 again, the calculations a lot of times the
6 calculations if not -- the only reason the
7 calculations are made I guess to ascertain
8 whether the magnitude is high enough to see if
9 it's reportable, it becomes a big issue. As a
10 process engineer, you know, it's not something
11 that I like to factor in and say, Well, that's
12 just normal practice, lost production.

13 Q. So, since DuPont was having trouble
14 locating the holes and cracks in the equipment
15 that were causing these external leaks they
16 really weren't in a position to calculate the
17 amount of gas that was coming out of those holes
18 and cracks, right?

19 A. Again, they can always calculate.
20 Somebody can always calculate.

21 Q. I'm sorry, they weren't in a position to
22 accurately calculate the amount of gas coming out
23 of the holes and cracks?

24 A. No.



1 Q. Do you know now whether -- have you seen
2 now any accurate documentation of the size of the
3 holes and cracks in the HIP, the CIP or the
4 converter?

5 A. No. I actually never been involved in an
6 incident investigation that helped ascertain the
7 -- the plant take it upon themselves to do the
8 investigation, assessment and calculations.

9 Q. We didn't cover this part, but you are no
10 longer with the Acid Technology Center, right?

11 A. That's correct.

12 Q. You transferred at some point to -- maybe
13 we did talk about that. They had a long
14 transition?

15 A. Right. Actually, I applied for a
16 transfer sometime in March of 2012. It was
17 approved, well, in principle by my then
18 supervisor and my current supervisor, but HR
19 works in mysterious ways and it wasn't officially
20 approved until October of 2012. But once there
21 was a mutual understanding between my current
22 group and my old group that I was going to leave
23 the group one way or the other I transferred
24 pretty much all my responsibility to Burnside and



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1 dedicated my time to my new group, which is
2 Global Engineer Solutions, still part of DuPont.
3 And the only involvement I had with Burnside
4 related to the replacement of the spent furnace
5 that I think they're going to install this month.

6 Q. And was your -- as you were applying for
7 a transfer to change positions was it a factor
8 that you weren't enjoying your working
9 relationship with the Burnside management too
10 much or the way they were handling things?

11 A. Yes.

12 Q. Would you have liked to see -- let me
13 strike that. Were you disappointed to see them
14 using this hose system for so long and relying on
15 that?

16 A. Yes.

17 Q. Did you have any conversations with Tom
18 Miller or anyone else in management at the plant
19 about that issue?

20 A. I had very limited discussions with Tom
21 Miller. In my initial discussions with them led
22 me to believe I would be wasting my time trying
23 to discuss anything related to the hoses.

24 Q. To the --



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1 A. To the hoses. I had made it known that
2 having the hoses for temporary solutions and,
3 again, temporary is a very broad term, right, but
4 I did not agree with the principles, but we still
5 were using the hoses. There is a number of
6 things. It was the working environment, the
7 operating principles of the plant in general. I
8 really learned enough of a new acid plant that I
9 saw no value to continue there. Even though I
10 think not only with Percy, but with the operators
11 in general I had a very good relationship, but
12 that was enough.

13 Q. So, unlike Red Lion you were treated more
14 like an outsider by the supervisors and the
15 management?

16 A. Yeah, again, I was an outsider. I don't
17 report to the plant. So, technically I'm an
18 outsider and I tried not to compare too much
19 actually to Red Lion. I knew it was a unique
20 situation, but in general this was not -- I
21 didn't see it worthwhile for me to continue.

22 Q. During this time period where you were
23 working with the Burnside folks did you get to
24 know Jeff Simoneaux, one of the operators?



1 A. Yes.

2 Q. And maybe Drew Tabor, did you get to know
3 Drew?

4 A. Yeah. I mean, I knew -- I met all the
5 operators, right, and some operators were more
6 friendly than the other ones; but, I think there
7 was I would like to say it was a mutual respect.
8 I respect what they do, what they're trained to
9 do, their responsibilities, and they respect I
10 think what I did because I wasn't trying to
11 establish a hierarchy between engineers,
12 operators.

13 Q. You got that mutual respect with the
14 operators, but maybe there wasn't so much mutual
15 respect with the plant manager, is that fair?

16 A. Yes, I would say that's fair. I can't
17 pinpoint to anything that was overly -- DuPont
18 doesn't tolerate lack of respect so there is
19 nothing overly.

20 Q. Overtly?

21 A. Overtly disrespect. Just, again, just
22 opinion, personal feeling, interactions that it
23 becomes evident at least to me.

24 Q. Do you recall -- and you may not remember



1 the date, but you are pretty good with dates, so,
2 you might -- do you recall around February 1st of
3 2012 being involved in some discussions with some
4 of the operators about Jeff Simoneaux reporting a
5 gas leak and maybe the plant manager's reaction
6 to that?

7 A. Yeah, yeah, I don't remember exact day.
8 I remember the event where I go to the plant and
9 Jeff and others described the situation to me and
10 I -- again, it took me aback what they were
11 describing, but I can't say -- I wasn't present
12 -- I wasn't at the plant at the time that it took
13 place.

14 Q. How did you first hear about it? Did
15 somebody call you?

16 A. No, no, no. Either I was there that week
17 or I just came. Again, in that period of time I
18 was still -- February 2012 -- I still had some
19 responsibilities and I probably haven't applied
20 for my new job. It hasn't been approved yet.
21 So, I just happened to be there and, I mean,
22 usually the first thing I do after dropping off
23 my backpack in the office is go to the control
24 room and that's when I was told.



1 Again, I don't think it was something
2 that they called me specifically about. Percy
3 and others have called me about situations before
4 when I'm not at the location. So, if you know
5 the date somebody probably told you that I was
6 there and I was told specifically, yes.

7 Q. Do you remember calling some of the
8 operators who had been there the night of this
9 incident?

10 A. Yeah, because I think Jeff told me about
11 it first. So -- again, it was just incredible
12 the reaction that Jeff described that Tom Miller
13 had taken. I asked him who else was there. I
14 don't exactly remember who the other operators
15 were, but I got verification that, yeah, that was
16 the case. Apparently Tom refused to go outside
17 to take a look at a leak. Is that the situation?

18 Q. That's what I think is written in this
19 log. This was the logbook entry for
20 February 1st. I think some of it was written by
21 Jeff himself.

22 A. Yes, yes. Yes, again, I do remember the
23 day, the situation, collaborating with other I
24 forget -- his operator partner, that, in fact,



1 they witnessed that and I will say that was
2 another nail in the coffin that my tenure at
3 Burnside was overdue.

4 Q. You were able to verify with the other
5 operators who were working that night that what
6 Jeff said had taken place with Tom Miller had
7 actually occurred?

8 A. I was able to -- they were able to
9 collaborate. I cannot verify that I was there
10 and heard what Tom said, so, yeah, the other
11 operators collaborated. Again, I know Jeff and
12 the other operators that I had -- I believe what
13 they said. I was not -- I didn't feel
14 comfortable enough to do what I would have
15 normally done. I didn't feel comfortable enough
16 to ask Tom whether that, in fact, took place.

17 Q. You didn't feel like you had a good
18 enough working relationship to approach him about
19 that?

20 A. Yes. Again, normally that would -- if
21 it's somebody I know or I did not believe that he
22 could say something like that, you know, maybe
23 it's somebody I know who said it and just in the
24 heat of the moment I would just, Did you really



1 say that? But, you know, by then it's obvious
2 that there was -- I felt an outsider, why would
3 he -- I had already tried from the inception of
4 when he was brought into the plant 2011, I think,
5 the turnaround, which is unusual to bring in a
6 plant manager during a turnaround. I tried to
7 establish a relationship, introduced myself, and
8 it was obvious that it was only one side, so.

9 Q. Meaning he was going to say how it is and
10 that's that and --

11 A. Meaning that -- meaning that whatever he
12 had to say at that time I was not interested.

13 Q. There was discussion or part of the issue
14 that Jeff wrote about in this log entry was that
15 he had started to cut back the plant when he saw
16 this leak, he started to cut back the rates of
17 the plant, and he has testified that Elizabeth
18 Cromwell had agreed with his recommendation to do
19 that.

20 A. Okay.

21 Q. But then Tom Miller arrived and overrode
22 that decision and said not to cut back the rates.
23 Do you recall anything about that, whether they
24 -- whether the rates kept going at the regular



1 rate?

2 A. No. It can be verified by the DCS very
3 easily. That part of the conversation I don't
4 recall. I would say it would be a normal
5 practice to if you have a leak to cut back on
6 rates because that reduces the internal pressure.
7 So, that minimizes. That doesn't get rid of the
8 leak, but it reduces the leak until somebody can
9 ascertain whether the plant is to be shut down or
10 continue running. So, if Jeff did that, that's
11 an excellent practice. If he checked with
12 Elizabeth, that was the normal thing to do.

13 Q. Have you ever known a plant manager to
14 override a recommendation by an operator to cut
15 back the rates when there was a leak or have you
16 ever been presented with that situation?

17 A. I haven't been presented with that
18 situation. I mean, a plant manager has the
19 authority on the plant. Again, depending on
20 circumstances, yeah, he can override it if a leak
21 is not big, but there has to be a reason; but, I
22 never firsthand witness that so I can't say it
23 never happens.

24 Q. Did you hear anything from the operators



1 that after that event they felt uncertain about
2 how to report or whether to report?

3 A. Yes, yes.

4 Q. Were you ever in your role with Burnside
5 ever asked to do any calculations on gas leaks?

6 A. By?

7 Q. External gas leaks.

8 A. Not directly. The spread sheet that I
9 showed you was -- I don't think Kerry Long was
10 management. It was the environmental guy and he
11 just wanted me to check his calculations or check
12 the parameters that were used to make the
13 calculations. Burnside management per se never
14 asked me to do calculations as part of incident
15 reports.

16 Q. You obviously would have been capable of
17 doing that had you been asked, right?

18 A. Incapable you say?

19 Q. You would have been capable.

20 A. Well, anybody can calculate. I would
21 have done my best to ascertain the conditions and
22 parameters where I can use those as a basis to
23 make the calculations.

24 Q. You would have pushed pretty hard to try



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1 to get an accurate size of the holes and cracks
2 and maybe duration of some sort?

3 A. I would have tried -- I wouldn't call it
4 push hard. I would have said, Dan, now, what are
5 the facts knowing there are a lot of assumptions.
6 One of the things I learned as an engineer you
7 don't have all the information all the time or
8 you don't have all the information every time.
9 Sometimes you have to make assumptions and what I
10 would like to do is state those assumptions
11 clearly so that the people that are involved in
12 the investigation are able to report either
13 internally or externally that these are the bases
14 and these are the uncertainties. I would
15 challenge anybody to be able to measure
16 accurately any type of leak such as from a duct.

17 Q. Without knowing the size of the cracks
18 and holes?

19 A. Right. I mean, you know, at Red Lion I
20 was involved in calculations because I was asked
21 to do that. I was part of the incident
22 investigations but, again, it's not something
23 that is -- I can force the plant and say you must
24 -- the calculations must come out from a document



1 that I issue. It's up to the plant to decide.

2 Q. Let's look back at the documents you
3 brought with you. I guess we can start with
4 Exhibit 1. Tell me about this document. You
5 were trying to do that earlier and I made you
6 wait.

7 A. Oh, okay, this is we -- based on
8 literature reviews that we had done this is
9 actually the best correlation that we can come up
10 with that we can agree in the industry of how to
11 determine gas leaks from equipment or from ducts
12 or piping. It makes certain assumptions with
13 regards to the ideality of how gas behaves, a
14 real gas, and the parameters that come into play
15 for the calculation are the difference in
16 pressures between the process container, whether
17 it's a vessel or a duct, and the ambient
18 pressure. Takes into consideration the physical
19 characteristics of the gas, composition,
20 temperature.

21 And, so, those are physical
22 parameters. Process conditions can either
23 measure fairly accurately or estimate it. I
24 discussed earlier the calculation makes -- it



1 estimates the mass of either S02 or S03 that is
2 released per unit time and the highest
3 uncertainty parameter is the area of the hole
4 where it's discharging.

5 The second calculation that is
6 typically asked for is how much, the quantity,
7 the mass of the other component that is in
8 question, S02, S03. So, the second uncertain
9 parameter is the duration. So, this is a typical
10 -- this is a spread sheet in probably pretty much
11 all the acid plants in DuPont has a similar
12 spread sheet that is based on this equation.
13 It's used as a basis to estimate leakage rate and
14 total amount of leakage.

15 Q. So, is this sort of a sample calculation
16 or is this reflecting a particular event?

17 A. This was a particular event where there
18 was a leak out of the duct exiting the 1st pass
19 converter.

20 Q. Do you remember when this was done?

21 THE WITNESS: I told you yesterday,
22 right?

23 MS. WEINER: I can't remember.

24 A. This was done in 2010.



1 MS. WEINER: He is referring to
2 Exhibit 2 now.

3 THE WITNESS: No, no, Exhibit 1.

4 MS. WEINER: Oh, okay.

5 A. Is it important?

6 Q. The precise date?

7 A. Yes.

8 Q. If you know it, your best estimate is
9 fine. If there is a way to determine it, you can
10 tell us that too.

11 MS. WEINER: We can probably get you
12 that after the deposition because you can check
13 on your computer.

14 THE WITNESS: Actually, the file name
15 shows the date. I know it was 2010.

16 MS. BARNEY: That would be fine.

17 A. So, this was -- at the time I had this
18 calculation Kerry Long had asked me to check some
19 of the parameters he was using as far as
20 composition, pressures and temperatures and some
21 of the constants and coefficients that were used
22 in the calculation.

23 Q. Do you remember which parameters he was
24 asking you to check?



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1 A. All of them.

2 Q. Kind of the whole --

3 A. Pressure, temperatures, composition of
4 the gas in the duct, exiting the 1st pass.

5 Q. He plugged in for you, I guess, the area
6 of the hole that is on here?

7 A. Yes, yes.

8 Q. And then he plugged in a duration?

9 A. Either he plugged it in or I asked him
10 what the duration would be. I mean, my effort
11 was to make sure he understood what the process
12 parameters, temperature, pressure, composition,
13 that he was using the right ones, because, as I
14 noted earlier, different ducts have different
15 compositions, pressures and temperatures. I was
16 able to pull up actual temperatures or estimated
17 pressures of that duct for the time in question.

18 Q. And you can do that from your computer in
19 Wilmington?

20 A. Right, process the data I can determine.
21 Cross section area of the hole, duration --

22 Q. That has to come from the plant?

23 A. That has to come from an incident
24 investigation, whether it's just directed by the



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1 plant or outside.

2 Q. So, for this particular calculation the
3 area of the hole -- well, the equivalent diameter
4 of the hole was a quarter of an inch, is that
5 right?

6 A. Right, because it's very hard to estimate
7 an area of an irregular shape, so.

8 Q. So, the equivalent diameter is when you
9 take the size and sort of convert it to an
10 equivalent -- it's the best you can do to
11 estimate the diameter?

12 A. The area of a circle is pi times the
13 square of the diameter divided by four. So, one
14 starts by saying I have seen the cross section
15 area is this much and then you can perform a
16 calculation and based on an assumed area you can
17 calculate the diameter. Or, somebody looks at
18 the hole and it's irregular and the largest
19 diameter is one inch and the smaller one is a
20 quarter, I think it's half an inch equivalent.
21 It is what it is, it's an estimate.

22 Now you can understand why I say this
23 is so uncertain trying to guess an area, but,
24 again, sometimes you have to make an educated



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1 guess.

2 Q. And that's just almost a conversion?

3 A. Right.

4 Q. Sort of converting an odd shape into a
5 circle?

6 A. Right, exactly.

7 Q. You try not to lose too much in the
8 process?

9 A. Right.

10 Q. But you are starting with a general
11 understanding of the size of the hole you are
12 looking at?

13 A. Right. The general understanding is the
14 -- the assumption in this equation, no matter the
15 shape of the hole, perfect circle, a square, a
16 star, completely irregular, no matter the shape
17 as long as it has the same area it will give --
18 it will calculate a leakage rate based on that
19 equivalent area.

20 Q. So, in this example a hole with a
21 diameter -- let me go back to area. The area of
22 the hole in this example was 0.00034 square feet?

23 A. Yes.

24 Q. And, so, that size area of a hole given



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1 all the other process parameters here,
2 temperature, pressure, generated a leak of SO₃ of
3 145 pounds for a duration of 86,400 seconds?

4 A. The equation give a leakage rate of
5 .0088 pounds per second.

6 Q. Where is that on here?

7 A. Mass calculation.

8 Q. I see, okay, Q equals?

9 A. Right, that is the leakage rate --

10 Q. Per second?

11 A. -- per second of total gas. And based on
12 that total mass flow and based on the duration
13 for the release, again, 86,400 seconds, which
14 translates to, I don't know how many hours.
15 Based on the composition of the gas having 19
16 weight percent SO₃, five weight percent SO₂
17 that's how we arrive to a total release of
18 145 pounds of SO₃ and 38 pounds of SO₂.

19 Q. You can use this calculator if you like,
20 but I ran that 86,400 seconds and if you divide
21 that by 60 --

22 A. Yes.

23 Q. -- you get how many minutes, is that
24 right?



1 A. Right. And you divide it by 60 again.

2 Q. I'll let you do it if you can use that
3 phone.

4 A. 1,440 hours -- excuse me, minutes,
5 minutes and 24 hours.

6 Q. This one was done on a 24-hour period?

7 A. Yes. So, that you can see it was either
8 conservative estimate or maybe not so
9 conservative. I have no way to tell without
10 somebody describing the scenario that, okay, this
11 is a basis of duration for and which will
12 determine the total amount release.

13 Q. Maybe they just were doing a per day,
14 they just picked a day.

15 A. Exactly, and I just want to make -- you
16 know, things like what Kerry Long will know is
17 composition of the gas because he has to
18 understand how much SO₂ got converted to SO₃
19 after the 1st pass. He might be able to get the
20 pressure and temperature from the DCS. The
21 temperature is directly -- is evident, but the
22 pressure has to be estimated because there is no
23 pressure meter right there. So, it wouldn't be a
24 case that I was involved -- this may not be the



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1 values reported.

2 The objective of this exercise, as I
3 say, is to make sure this is the right
4 parameters, process parameters, by which he
5 should use to make the calculation.

6 Q. Can you calculate -- if I gave you a
7 different area and everything on here stayed the
8 same, can you estimate what the increase in
9 pounds would be or is that something you --

10 A. Depending on -- do you want to double the
11 area?

12 Q. Let me see. Yes, why don't we first try
13 doubling the area.

14 A. Then double the rate, it would be .0176.
15 You can see that the leakage rate is directly
16 proportional to area. So, you double the area,
17 you double the leakage rate.

18 Q. And the leakage rate is Q on here, right?

19 A. Q. It's in the mass -- again, that is
20 the mass of the total gas. There will be some
21 oxygen in there, some nitrogen in there, some
22 CO2.

23 Q. So, if you -- the percentage of that
24 doubled rate that is S03 how do we arrive at



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1 that?

2 A. Again, because we didn't change the
3 concentration --

4 Q. We just doubled.

5 A. You double the flow, same concentration,
6 you double the leakage rate for S03, which would
7 be 290 -- if you maintained the same duration,
8 24 hours, it would be 290 pounds of S03 and
9 76 pounds of S02.

10 Q. If the area of the hole was twice what it
11 is on this sheet right here?

12 A. Right, right. I'm glad you didn't say
13 2.3 higher. Then, I can't do the math in my
14 head.

15 Q. Can you tell me about what the area would
16 be for a hole that is eight inches by 1.16?

17 A. Eight inches by 1/16? It would be .15
18 square inches.

19 Q. And that if we -- let's see. So, you
20 would divide I guess by the area on this page?

21 A. No. It would be -- this area is in
22 square feet, so, it would be .00345 square feet.

23 Q. So, about ten times --

24 A. Yes. Let me verify that since this is



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1 not a familiar calculator. I didn't realize I
2 had to do calculations, so.

3 Q. I didn't either.

4 A. You say eight inches, right?

5 Q. Yes. By 1/16.

6 A. .00347, so, about ten times.

7 Q. A hole that size or a crack that size
8 would have an area that is about ten times the
9 area that was used on Exhibit 1?

10 A. Right, if only thing that changed was the
11 hole size.

12 Q. If all the other conditions were the
13 same?

14 A. Right, because, again, pressure,
15 temperatures, pressure in particular, the
16 differential pressure between the internal duct
17 and the ambient has a big effect on leakage rate.

18 Q. So, if the area reflected by eight inches
19 by 1/16-inch crack, if that area is ten times the
20 area that was used in the Exhibit 1 calculation,
21 then the total gas amount where it says Q on
22 Exhibit 1 would also be ten times that?

23 A. Right.

24 Q. And then the S03 mass calculation would



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1 be ten times the 145 pounds?

2 A. Right, if everything else remains the
3 same.

4 Q. So, if everything else was the same an
5 eight inch by 1/16-inch crack would generate
6 under these conditions in Exhibit 1 1450 pounds
7 per hour --

8 A. Yes.

9 Q. -- of gas leak?

10 A. Of S03 would be released in that 24-hour
11 period. I think you wrote pounds per hour.

12 Q. Oh, I did. What is it supposed to be?

13 A. Total pounds for the duration, which was
14 24 hours for a 24-hour hour period.

15 Q. 1450 pounds for a 24-hour period and
16 that's of S03?

17 A. Right.

18 Q. Where on Exhibit 1 does it tell you the
19 rate at which the plant was running or the
20 assumption that was made about the rate at which
21 the plant was running when this calculation was
22 done?

23 A. Because the rate would not have -- the
24 production capacity will not be part of the



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1 calculation directly. Indirectly it does. The
2 higher the production the higher the pressure in
3 the process duct. So, even though it doesn't
4 show -- the higher the production the equipment
5 has to work harder to push more gas, pressure
6 goes up, leakage up.

7 That's why, again, Jeff did the right
8 thing or any operator did the right thing if
9 there is a leak and if you don't have the
10 authority or it's not bad enough to shut the
11 plant because shutting down the plant -- I mean,
12 restarting it could cause more damage. The right
13 thing to do is slow it down. By slowing down the
14 rates you reduce the pressure, you reduce the
15 leak, the leakage rate.

16 Q. Where on Exhibit 1 do you see the
17 pressure reflected?

18 A. Under P source pressure absolute and the
19 equation is the third parameter and he also
20 factors into the denominator inside the equation.
21 So, there are three places where P plays a part,
22 but you can see the pressure is also directly
23 proportional.

24 Q. To?



1 A. To the leakage rate. In this case
2 doubling the pressure does not necessarily double
3 the rate. You have to go through the whole
4 calculation to determine that. In fact, I won't
5 --

6 Q. I'm not going to make you do that with my
7 cell phone.

8 A. I can guess what it's going to be, but
9 Monique may not be happy with me guessing.
10 Production has an effect. It's not a number that
11 is plugged into the calculations, but it's
12 reflected in pressure.

13 Q. Is there a rough estimate that you know
14 of the proportion between the increase in
15 pressure and the amount of SO₃ gas or is that not
16 --

17 A. No. Pressure is a function of production
18 rate. SO₃ concentration is a function of which
19 duct is leaking, right?

20 Q. So, on this calculation do you have a
21 rough estimate of if you increase the pressure on
22 this calculation what it does to the SO₃ mass
23 calculation?

24 A. The total mass?



1 Q. Yes.

2 A. Yes, if you -- again, if you increase the
3 pressure, you increase the leakage rate and since
4 the leakage -- even though the concentration of
5 S03 hasn't changed because you increase the
6 leakage rate you are going to increase the amount
7 of S03.

8 Q. By looking at this pressure does that
9 tell you under normal plant operating conditions
10 about what rate the plant was operating at?

11 A. If I was more in tuned to the plant, yes,
12 but, again, there is a correlation. By looking
13 at it I can't tell. I mean, to me the easiest
14 way is I can pull that data directly from the DCS
15 what that production rate was at that particular
16 time. I don't need to try to guess.

17 Q. There is something on the computer where
18 we could see every day for the last two years
19 what the production rate was and what the
20 pressure was and what the temperature was?

21 A. For the production, yes. For the
22 pressure there is only a limited number of
23 pressure transmitters, recorders, throughout the
24 plant. So, that's why for this particular



1 situation I had to estimate the pressure in that
2 duct based on two pressure transmitters upstream
3 and downstream, estimate pressure losses in
4 between and what is the reasonable pressure in
5 that location.

6 To confirm it, again, based on those
7 estimates we do monthly field surveys of what the
8 pressure is in not exactly that location, but
9 very near; and, so, I can correlate whether the
10 estimated pressure at that particular point is
11 reasonably accurate.

12 MS. BARNEY: And, Monique, I think we
13 requested any data that went into calculations
14 like this. I think that was one of my requests
15 for production. So, I would just ask if you
16 could check on producing the -- is it the D --

17 THE WITNESS: DCS data.

18 MS. WEINER: I will look into, first
19 of all, whether it was asked for in the first
20 place, I will look into it.

21 THE WITNESS: DCS is data collection
22 system.

23
24



EXAMINATION

BY MS. BARNEY:

Q. All you remember Mr. Long asking you is just to confirm the parameters that you already talked about?

A. Yeah, the process parameters.

Q. He didn't say anything about why he was doing the calculation in the first place that you recall?

A. Well, he didn't have to ask me this. It's a leak out of the 1st pass outlet.

Q. And that was in 2010 before these other leaks really started, right? This must just have been another leak?

A. I think this was early when I start supporting Burnside.

Q. If there was a leak back then it must have been one that they were sort of able to get under control because, as I understand it, the leaks that they couldn't keep under control or get repaired started in December 2011, is that right?

A. Okay.

Q. They might have started earlier I guess



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1 is what --

2 A. Yeah, I'm not sure of the exact dates. I
3 will say the leaks in itself is not something
4 that they would communicate to me for particular
5 reason that I wouldn't be the specialist to help
6 them fix it.

7 THE WITNESS: I'm going to step to
8 the restroom.

9 (Brief recess.)

10 EXAMINATION

11 BY MS. BARNEY:

12 Q. Let me show you a document that we'll
13 mark as Exhibit 5.

14 (Chu Deposition Exhibit No. 5 was
15 marked for identification.)

16 EXAMINATION

17 BY MS. BARNEY:

18 Q. This is Bates labeled DSF0000083. I ask
19 you to look at that. Is that basically the same
20 calculation type that we were looking at in
21 Exhibit 1?

22 A. Well, technically I will say no. It's
23 missing the molecular weight of the gas. If you
24 compare Exhibit 1 and 5.



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1 Q. Exhibit 5 is missing the molecular
2 weight?

3 A. Right, but that's not to say that the
4 calculation is wrong. For example, this formula
5 --

6 Q. At the top?

7 A. -- at the top of Exhibit 5 is just -- it
8 doesn't play into the calculation itself. It's
9 just to show this is the formula that has been
10 used just like I have it here.

11 Q. On Exhibit 1?

12 A. Right. I know this is not -- in
13 Exhibit 1 this item is just an image. The
14 formula used here to do the calculations -- I
15 mean, the calculation is based on this formula.

16 Q. In Exhibit 1?

17 A. Right.

18 Q. Okay.

19 A. So, I would have to either run the same
20 calculation or look at the original spread sheet
21 and check the formula that is in that cell or do
22 the calculation manually on a calculator with all
23 these parameters to see if it gets that. So --

24 Q. And that's for Exhibit 5?



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1 A. Exhibit 5. So, the result that is shown
2 I don't know if it's using the right formula
3 because the represented formula does not match
4 the one that is supposed to be in there.

5 Q. So, the formula that should be the basis
6 for the calculation on Exhibit 5 is not the
7 formula that is at the top of Exhibit 5?

8 A. Right.

9 Q. The formula on Exhibit 1 in the yellow
10 box --

11 A. It's missing the molecular weight of the
12 gas.

13 Q. I see an M in the first set of
14 parentheses on Exhibit 1 in that formula.

15 A. Right.

16 Q. But on Exhibit 5 the formula doesn't have
17 an M in it?

18 A. Right. But, again, it doesn't mean that
19 the calculation is wrong. It's just when they
20 cut and paste from whatever document it was
21 pasted from, that image, that formula was wrong
22 because I do notice in Exhibit 5 molecular
23 weight, M, is shown as a parameter.

24 Q. Right.



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1 A. But, again, visually looking at -- I just
2 print the document. I cannot determine whether
3 that calculation is done right or not.

4 Q. They might have plugged in a value for
5 molecular weight of gas, but when they crunched
6 the formula if they followed the formula at the
7 top of Exhibit 5, the calculation would have not
8 have reflected the molecular weight of gas?

9 A. Right, but, again, on the other hand the
10 actual formula might have taken that into
11 consideration and it's correct.

12 Q. If the Excel spread sheet did it by
13 itself using the right formula, then it would be
14 captured?

15 A. Right, right. This is 2nd pass. Um, I
16 would say in general the parameters being used
17 are I would say is good, I mean, the composition,
18 temperatures, pressures are consistent with the
19 2nd pass exit.

20 Q. Have you ever seen this particular
21 calculation before, this page reflecting this
22 calculation?

23 A. This particular page probably not. I'm
24 trying to determine whether -- this was obviously



1 they had gotten this from Burnside -- whether I
2 got this and modify it and gave it back to Kerry
3 with the correct formula.

4 Q. You might have received Exhibit 5 and
5 given him the right formula and sent it back, but
6 then do you have a recollection of that, changes
7 of it from 1st pass to 2nd pass only?

8 A. No. That would be done by him or
9 somebody else. But, again, just looking --
10 again, without calculating the numbers the
11 process parameters are slightly different, but
12 the big difference is the equivalent diameter of
13 the hole is three times bigger, right?

14 Q. Oh, okay.

15 A. And it shows a leakage rate that is
16 almost nine times because it's .0801 pounds per
17 second.

18 MS. WEINER: Which one are you saying
19 shows the nine times, Exhibit 1 or Exhibit 5?

20 THE WITNESS: Exhibit 5.

21 A. Excuse me, let me backtrack a little bit.
22 The equivalent area of the hole is ten times
23 larger. Exhibit 5 is .00323. Exhibit 1 is
24 .00034. So, we have established that the leakage



1 rate is directly proportional to area. So, you
2 have ten times the area, you have ten times the
3 leakage rate. So, even though the pressures are
4 slightly different, temperatures are slightly
5 different, the biggest impact on the leakage rate
6 is the area.

7 So, again, without doing all the
8 calculation it's using the right formula. The
9 representation -- the image of that
10 representation on Exhibit 5 is just wrong. Does
11 it make sense?

12 Q. Yes. So, using the calculation on
13 Exhibit 5 the release duration for Exhibit 5 is
14 --

15 A. Is different.

16 Q. -- smaller?

17 A. Right.

18 Q. So, the total mass -- if we were going to
19 calculate the total mass for Exhibit 5 based on
20 -- can you tell me how many hours the duration is
21 for Exhibit 5?

22 A. I just want to make sure it's understood
23 that the formula calculates only leakage rates,
24 pounds per second. So, looking at a consistency



1 check and neglecting the difference in process
2 conditions the main difference between Exhibit 5
3 and Exhibit 1 for leakage rate calculation is the
4 equivalent area of the hole. So, if Exhibit 5
5 was ten times bigger than Exhibit 1, the leakage
6 rate should be about ten times bigger, which it
7 is, .0801 pounds per second versus .0088, leads
8 me to believe that Exhibit 5 used the correct
9 formula. It just cut and pasted a wrong image.
10 So, the next question is what was the duration of
11 that leakage?

12 Q. Yes, if you can just crunch the number of
13 seconds that is used on page 83.

14 A. I'll use my calculator because you just
15 locked me out. Actually, I'll use an actual
16 calculator. Exhibit 5 has a duration of 35,496
17 seconds, which is equivalent to 9.86 hours.
18 Okay, 9.86 hours.

19 Q. So, on Exhibit 5 using the area of the
20 hole on that exhibit they got a total release in
21 you said about nine hours?

22 A. Actually, closer to ten, ten hours.

23 Q. Ten hours, of 318 pounds, is that right?

24 A. 318 pounds of S03, yes.



1 Q. One question I had is I see where you are
2 saying the area of the hole in Exhibit 5 where it
3 says A equals area of the hole?

4 A. Mm-hmm.

5 Q. That appears to be ten times the A value
6 area of hole on Exhibit 1, but the figure that is
7 beside the line that says equivalent diameter of
8 hole on Exhibit 5 doesn't seem to be ten times
9 bigger than the line that says equivalent
10 diameter hole on Exhibit 1.

11 A. Because area is proportion to the square
12 of the diameter. So, even if you -- let's say
13 you doubled the diameter, you quadruple the area.

14 Q. Okay, we should go by the A value, the
15 area of the hole --

16 A. Right, right.

17 Q. -- and not by the line that says
18 equivalent?

19 A. Again, they assume a .77 diameter in
20 calculating the area, but you can do the reverse.
21 I just checked the math and the math is correct.
22 Again, this is not -- twice -- so, you can
23 imagine a small hole you double the diameter, you
24 actually quadruple the area.



1 Q. Could you maybe just so the record is
2 clear would you mind circling the part of
3 Exhibit 5 that you think is that formula on there
4 and you can just circle that it's missing the M.

5 A. (Witness complies.)

6 Q. It appears the calculation was done using
7 the right formula?

8 A. The formula that is built into the spread
9 sheet that is used for the calculation appears to
10 be correct.

11 Q. If you looks at Exhibit 5 you see beside
12 that 318 pounds it says, "Note: S03 immediately
13 reacts with H2O." Do you know what field in the
14 spread sheet that would be written in? Is that
15 just a field for comments?

16 A. It's just a field for comments.

17 Q. Do you have any idea who wrote that?

18 A. No. But, again, that's a common -- S03
19 will react with water to make sulfuric acid mist.

20 Q. Is it your expertise to determine what
21 the reportable quantity is for a certain chemical
22 under a particular statute?

23 A. No. Reportable quantity is determined by
24 the statutes. Permits in each state I think is



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1 different. I'm not sure if the federal
2 standardized it for each state.

3 Q. I'll just show you these and see if you
4 can tell me what these documents are. They're
5 Bates labeled 315 to 318.

6 MS. BARNEY: We can call it
7 Exhibit 6.

8 (Chu Deposition Exhibit No. 6 was
9 marked for identification.)

10 EXAMINATION

11 BY MS. BARNEY:

12 Q. First page is 315 and I'm just wondering
13 if you can tell me what that document is.

14 A. This is a drawing of the converter for
15 Burnside. It's original drawing that reflects a
16 single absorption configuration.

17 Q. So, that's before they moved to the --

18 A. Dual absorption.

19 Q. This really does not reflect the
20 equipment that was in use starting in
21 December 2011?

22 A. The converter itself is correctly
23 reflected. For example, the ducts that connects
24 the exit of 2nd pass into the 3rd pass and the



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1 exit of the 3rd pass and the 4th pass indicates
2 to me that it was single absorption and that
3 would have been changed; but, the converter
4 itself, the size, the dimensions, are the current
5 one.

6 Q. Can you maybe just put a mark where you
7 think this document does not reflect the current
8 state of the converter?

9 A. (Witness complies.)

10 Q. And the second page, block flow diagram?

11 A. This is a design to provide a simple
12 representation of the Burnside plant, simple
13 process representation of the Burnside plant, and
14 I drew this. Maybe I shouldn't have admitted
15 that.

16 Q. No, I like it. The CIP and the HIP and
17 the converter are sort of at the bottom towards
18 the left of the page, right, that's where they're
19 shown?

20 A. Right. Because it's a simplified -- it's
21 designed to represent the major inlets and
22 outlets of the process, not to truly represent
23 how each line is connected, yes, that's correct.

24 Q. When do you think you drew this? After



1 you started working with the Burnside site?

2 A. Yes, I would say around 2010. I mean,
3 the intention was actually a teaching tool for
4 the new engineer and to the operators. Even
5 though the operators knew how to operate the
6 plant sometimes I will draw this, a simplified
7 drawing, and highlight particular inputs and
8 outputs.

9 Q. What is the third and fourth pages?
10 Those are pages 317 and 318.

11 A. The 317 represents three Monplx modules
12 that comprise the CIP and the page 318 consists
13 of the three modules that form the HIP. So, just
14 mechanical drawings showing dimensions, not
15 internals of the heat exchangers.

16 Q. By the time of your involvement with
17 Burnside did you learn anything about the actual
18 cracks and holes in the equipment?

19 A. No, not really. Again, those pertain to
20 external leaks. My focus was primarily to help
21 identify the onset of internal leaks.

22 Q. And you wouldn't have any information
23 about what was done in a recent turnaround, like
24 in September, October, of this year?



1 A. No.

2 Q. Other than maybe just talking with
3 people?

4 A. Other than talking with Percy, yeah. I
5 know they tried to find the leaks and they didn't
6 find them.

7 Q. And that's even in this most recent
8 turnaround?

9 A. Yes. But, again, not a lot of details,
10 no.

11 Q. Just sort of in passing because it's not
12 your job so much?

13 A. Well, it's not it's not my job. It's
14 hard to determine -- you know, it's just
15 information, Oh, you have a leak. It was nothing
16 I could contribute because even if it's not my
17 job it's still part -- it's a DuPont plant and
18 even Red Lion I had no responsibilities, but the
19 current plant manager used to be operations
20 manager when I was there. So, he calls me and I
21 respond to him as a friend.

22 Q. And you still care?

23 A. I still care. I mean, I had gone to Red
24 Lion several times. Even if I care there is not



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1 enough information for me to contribute. So,
2 that's why it's just discussion in passing,
3 checking on Percy, number of hours he is working,
4 and he will relate a few things to me.

5 Q. And nobody else from DuPont Burnside has
6 called you to get your input into the situation
7 either during the turnaround or after, is that
8 right?

9 A. 2013?

10 Q. Right.

11 A. No.

12 Q. Has anybody shown you any pictures from
13 the turnaround of cracks or holes in the
14 equipment?

15 A. No.

16 Q. If you saw pictures from the turnaround
17 of the equipment and damage to the equipment,
18 would you be able to tell if that was the source
19 of an external leak?

20 A. If I'm looking at equipment from the
21 outside, yes, I would tell it was an external
22 leak. Sometimes we do internal inspections. I
23 will say that black and white pictures are very
24 hard to do.



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1 Q. Yes, they're really bad. I'll just show
2 you these and see what they mean to you, if
3 anything.

4 A. With the understanding that I'm not an
5 expert on leaks.

6 Q. I understand.

7 A. Visual assessment.

8 Q. And I don't want you to say anybody you
9 are not confident about. This is DSF numbers 900
10 to 902. It's an e-mail from --

11 A. Mark Macha.

12 Q. -- Mark Macha to Gene Clemons with
13 attaching some photos.

14 A. Stripped off insulation.

15 Q. I think this must have been during a
16 repair attempt, I guess. This was right after
17 you left?

18 A. By May I would have been very little
19 association at Burnside. I'm trying to remember
20 when I went for the last time just to clear my
21 locker and move on. It's hard to -- not only
22 because it's not a color picture, but I can't
23 tell exactly which part of the CIP this picture
24 refers to. He indicates from the e-mail that he



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1 is stripping the insulation to see the external
2 surface of the CIP itself, but because they are
3 close-ups I don't have perspective of which side
4 the heat exchanger it involves.

5 Q. You can tell that this is the CIP, or,
6 that's the subject?

7 A. That's the subject. If I look at the
8 pictures without a reference I couldn't even tell
9 you -- well, probably I could tell you it's a
10 Monplx exchanger because it's flat, but I
11 wouldn't be able to tell you which one and even
12 with the reference I don't know the location.

13 Q. The only thing it would indicate to you I
14 guess is that the insulation was taken off and
15 they don't have insulation on the inside?

16 A. Right, and, again, because the contrast
17 in black and white too you can't tell whether by
18 discoloration alone which areas are corroded or
19 potentially could be leaking.

20 Q. If you see sulfates on equipment, what
21 does that tell you?

22 A. Well, that either there is a leak or
23 there is acid that drip on it. Sulfates by
24 itself, well, you know, are a result of acid



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1 dripping and eventually oxidizing into sulfates,
2 but, where you find it may or may not be the
3 source of where the leak originated.

4 Q. So, the leak is probably above the place
5 where you find sulfates?

6 A. Again, generally people that look for
7 leaks look for the signs where fumes come out,
8 sulfates accumulate, holes appear in the
9 insulation. So, you do have color pictures.

10 Q. I have some.

11 A. Things like maintenance, doing some
12 maintenance where some acid drip out of pipe and
13 ducts, eventually can lead into formation of
14 sulfates.

15 Q. I'll ask you to look at this picture and
16 I'll represent to you underneath the 3rd pass
17 duct, but I don't have -- does that look like
18 sulfates associated with a leak to you?

19 A. Again, it's hard to tell if it's
20 sulfates, wetted insulation or just corrosion
21 flakes from the steel. Again, it's not really --
22 again, my area of expertise is to not only --
23 leaks sometimes are obvious, but there are people
24 that can look at discoloration equipment in duct



1 and say this is the type of leak even by
2 discoloration, the temperature.

3 Q. Wow. Who can do that at DuPont?

4 A. Materials.

5 Q. A metallurgist?

6 A. A metallurgist, yes, and Burnside uses
7 various, but I know most of the good ones rely on
8 pictures if that's the only thing, but a lot of
9 times they have to visually see a condition, yes.

10 MS. BARNEY: Let's mark this.

11 (Chu Deposition Exhibit No. 7 was
12 marked for identification.)

13 EXAMINATION

14 BY MS. BARNEY:

15 Q. Do you remember adding a knockout pot to
16 the end of the CIP coming from the IPAT?

17 A. Yes, I actually call it a boot. That's
18 what we referenced earlier.

19 Q. Are you familiar with the forms that
20 DuPont Burnside called an Incident Investigation
21 Report?

22 A. Yes, I'm familiar with them.

23 Q. Did you ever have to participate in
24 putting those together at Burnside?



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1 A. No.

2 Q. That's right, you talked about that, you
3 weren't really asked to participate in that?

4 A. Right.

5 Q. You weren't asked, okay, but you did
6 participate in those at Red Lion?

7 A. Yes. Again, in my role is -- the way
8 incident reports work is either driven at the
9 plant level or at the business level and members
10 of the team are selected based on the expertise
11 needed.

12 Q. I'll show you two documents. One is
13 Bates labeled DSF15 through 19 and the other one
14 is 20 through 23.

15 MS. BARNEY: I guess we'll go ahead
16 and mark them so we can talk about them. Number
17 15 to 19 will be Exhibit 8 and 20 through 23 will
18 be Exhibit 9.

19 (Chu Deposition Exhibit Numbers 8 and
20 9, respectively, were marked for identification.)

21 EXAMINATION

22 BY MS. BARNEY:

23 Q. Exhibit 9 has a date on it I believe of
24 August 8th of '11.



1 A. Mm-hmm.

2 Q. Wait, Exhibit 9, sorry, has a date of
3 August 8, 2011 and Exhibit 8 has a date of --

4 A. June 11, 2012.

5 Q. Have you ever seen an investigation
6 report that had Burnside's name at the top?

7 A. Yes. Usually any incident report for any
8 of the acid plants, Burnside, Red Lion, they all
9 go to a wide distribution list. So, it is
10 accessible. I can't say that I read all of them.

11 Q. Do you recall getting any for the
12 Burnside site after June 11, 2012 or would you
13 have no longer been on the list at that point?

14 A. June 11th -- you say June 11, 2012,
15 right? No, I would no longer have been involved
16 even though technically I was still in ATC.

17 Q. You probably wouldn't have been on the
18 distribution list after that point?

19 A. Again, I probably was in the distribution
20 list. I get a bunch of e-mails, I look at the
21 heading and I choose to --

22 Q. Read or not read?

23 A. -- read or not read because there are too
24 many. If the incident report comes to Burnside



1 and says gas leak, chances are I would not have
2 read it because, again, it wasn't pertinent to my
3 role at that time. Usually there would be the
4 author -- usually the team members involved. I
5 just want to make sure I didn't perjure myself.
6 My name is not there.

7 Q. If you look at Exhibit 9 it seems to have
8 a spot for quantities or amount of release on the
9 first page?

10 A. Yes.

11 Q. And if you look at Exhibit 8 it doesn't
12 seem to have that. I was wondering if you knew
13 if the form changed or why that would be?

14 A. The form looks like it changed. I'm just
15 looking to see what it looks someplace else.
16 Obviously, it has changed from 2011 to 2012, but
17 I don't know why it was changed and why some
18 information was added and omitted.

19 Q. In order to get the number that would go
20 on a form that looks like Exhibit 9, the one that
21 asks for amount, would they have to do the
22 calculation that we talked about before on
23 Exhibit 1?

24 A. Hopefully they did a calculation. Again,



1 whether they used that spread sheet or something
2 similar or any other means I don't know. Looked
3 like they did a calculation. I can't tell what
4 they used.

5 Q. So, in order to put a number right here
6 they would need to have those same parameters we
7 talked about, area of the hole, duration,
8 pressure?

9 A. Yes, if they used that formula, yes, they
10 would have to get that information. Again, just
11 by the results I cannot tell whether -- what was
12 the basis of the calculation or what formula was
13 used.

14 Q. Were you asked or involved in the
15 calculation that is here for August 8th of 2011?
16 I think you may have already answered that by
17 saying you never did any of those at Burnside,
18 but I just want to be sure because you were
19 involved with them at that time.

20 A. My answer would be not that I recall. I
21 mean, a lot of calculations -- again, usually if
22 it's a calculation that I do I will provide in
23 written form with an e-mail so that -- again, so
24 that we have a paper trail so that if somebody



1 looks at the incident report and say, well, what
2 was the basis, we wouldn't have to guess the way
3 we are trying to do now.

4 But, again, to the best of my
5 recollection I wasn't involved in this
6 calculation and there are standard forms that the
7 plant could have used.

8 Q. To do the calculations?

9 A. To do the calculations.

10 Q. What form is that? Is that the form we
11 were talking about on Exhibit 1?

12 A. Right, the two exhibits, the one that --

13 Q. Exhibit 1 and Exhibit 5?

14 A. Right.

15 Q. If somebody at Burnside did the
16 calculations that we talked about on Exhibit 1
17 and 5, those would be saved on the computer
18 somewhere? Would that be the normal protocol?

19 A. It would be normal for me. Some people
20 don't like to save anything. Some people -- but,
21 the best way I can answer is for a calculation
22 like this where the interest is just a summary we
23 either have or should have hard copies or
24 electronic copies that you can insert as an



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1 attachment, again, to show what was the basis,
2 you know. A calculation is just an arithmetic
3 exercise. That's not the important part.

4 The important part is what
5 assumptions were used to arrive at the
6 calculation. So, it's best to document it. I'm
7 okay being wrong, but I don't want to distract
8 the fact that what I don't remember. So, again,
9 I can't tell what kind of calculation and what
10 parameters were used.

11 Q. If there were attachments inserted into
12 this document, Exhibit 9, would it be listed
13 under preliminary attachments? Would that be
14 reflected on this report?

15 A. Yes, it would be reflected. You will see
16 a little icon that says preliminary attachments
17 on Exhibit 9. There will be a little icon that
18 says a file was inserted there.

19 Q. And there aren't any reflected here on
20 page 20 of Exhibit 9?

21 A. Right, no.

22 Q. On page 20 of Exhibit 9 there is -- it
23 says chemical release and then it lists two types
24 of gas, right?



1 A. Yes.

2 Q. And under the amount released it only
3 lists one number?

4 A. Right.

5 Q. If you were doing that spread sheet we
6 talked about there should be two numbers, right,
7 one for each gas?

8 A. Yes, unless they totaled it. Right, the
9 calculation will discern two numbers whether the
10 amount reported or released reflects the sum of
11 those two.

12 Q. So, we just don't know I guess here
13 whether that was --

14 A. Yeah, I can't tell.

15 Q. This doesn't tell you the size of the
16 leak source, right, nothing on this page?

17 A. No.

18 Q. But it does indicate a duration?

19 A. Yes, 315 minutes.

20 Q. Then, with the form that is dated
21 June 11th, which is Exhibit 8, this form
22 discusses a gas leak from the CIP doesn't capture
23 the amount released -- I see now there is a place
24 for it down at the bottom of page 15 it says



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1 amount released?

2 A. Yes.

3 Q. But it's just left blank?

4 A. Yes.

5 Q. Normally -- let's see if this says it's
6 final. Can you tell from the last page? It says
7 complete issued?

8 A. The signature when it was drafted on
9 June 11, 2012 he has the progress, I would assert
10 that there would be a date for complete and a
11 date for the incident issue. That would be the
12 normal. Even though it says final maybe --
13 again, I don't know all the procedures whether
14 they -- but if it's complete in issue category
15 and it has a date in it, I would expect somebody
16 to say, okay, this is final and then the very
17 next step is it's complete and it was issued for
18 review like for the wide distribution list like
19 myself a particular date.

20 Q. So, we have it showing final from the
21 plant manager, but we don't have a date for when
22 it may have gone to the distribution if it did?

23 A. Right.

24 Q. Does this one show any attachment such as



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1 any spread sheet?

2 A. Let's see the date. Not that I can tell.
3 If you look at for both documents under status
4 edit tracking it shows the dates that they were
5 updated and ready to be issued. So, again, I
6 can't tell from this whether it officially issued
7 or there was another revision; but, as far as
8 that document it was updated to be ready to be
9 issued.

10 Q. Normally when you saw final investigation
11 reports were the blanks filled in for things like
12 amounts released out of primary container?

13 A. Normally, yes. I mean, that's a key
14 category to determine whether the release is
15 reportable, whether it was or was not, at least
16 there should be a basis for that.

17 Q. If you look on page 17 it says
18 environmental rating C, but they would need to
19 know the amount released in order to arrive at
20 that rating, right?

21 A. Again, on page 16 --

22 Q. I'm sorry, 17.

23 A. Okay, but I'm referring to page 16
24 environmental rating, line two, Actual size of



1 release, incident or magnitude of event, Small
2 release bigger than .45 kilograms, less than
3 4.5 kilograms. Again, I can't tell if this is an
4 example or the determination. The calculation
5 even though it's not shown fell between that
6 category and, therefore, because it put the
7 environmental rating with zero points, which
8 indicated very small release.

9 I do agree with you that the amount
10 of release is not stated, duration is not stated
11 in Exhibit 8, but I can't explain why it was not
12 included.

13 Q. If you look on page 15, Bates page 15, it
14 says the gas leak was traveling in the 5 and 6
15 tankcar rack area and we shut down tankcar
16 activity. For a leak that travels across the
17 site and requires clearing out a work area would
18 you expect to see a spread sheet done on the
19 quantity of the release?

20 A. I would expect the calculation if the
21 parameter is available allow for the calculation,
22 right. This is a CIP gas leak, so, it would have
23 come from the duct or the external housing of the
24 heat exchange. There would have been enough



1 parameters to estimate the leak.

2 Q. You mean because they could go get access
3 to the leak source by lifting up the insulation?

4 A. Yeah. I mean, if it was a leak -- again,
5 even if the leak is small, you know, even by
6 estimating the -- by lifting the insulation and
7 seeing it or because this is from the CIP and it
8 doesn't tell me which part of the CIP is leaking.
9 They can visually gauge the magnitude of a leak
10 based on the fumes, visible fumes.

11 They can still come up with an
12 estimate, so, they would have somehow come out
13 with an estimate, whether it's calculation or
14 just educated guess, to ascertain the magnitude,
15 small, medium or large.

16 Q. If you are not there in June of 2012 to
17 do an estimate, kind of a rough estimate I guess
18 on just visual appearance, then who would be
19 qualified to do that at Burnside in June 2012, if
20 you know?

21 A. Dan Monhollen, Elizabeth, Tom. It's a
22 spread sheet.

23 Q. The calculation?

24 A. Right, it's just a spread sheet. They



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1 can get the process data from the DCS. They can
2 get -- again, that's why the incident
3 investigation there will be discussions of size
4 of leaks based on the expected hole size or -- I
5 mean, if fumes get released, the duration.

6 Q. To accurately do it they have to do the
7 spread sheet calculation that we talked about
8 earlier and actually have an estimate of the area
9 of the hole and that sort of thing?

10 A. Again, the accuracy can be debatable, but
11 it's the basis that one uses, again,
12 understanding the uncertainty, if not duration,
13 the size of the hole. I don't associate
14 calculations with -- just because one can
15 calculate it and arrive at a particular number
16 that is accurate, but at least in the calculation
17 all the assumptions are put in place rather than
18 somebody just saying, looking, recollecting, and
19 saying I think that was 1.1 kilograms that
20 escaped. I don't know that anybody can do that.

21 Q. Do you recall being asked in May of 2012
22 -- so that would have been sort of when you were
23 transitioning out of Burnside -- for any
24 information that Burnside might have needed to



1 respond to an OSHA inquiry?

2 A. No, i don't recollect anything like that.

3 Q. I guess I'll show you a document that we
4 can mark as the next exhibit. That's an e-mail
5 from Jeff Simoneaux to Elizabeth Cromwell back on
6 February 1st. I think that was the incident we
7 were talking about earlier. I'll ask you just to
8 read that e-mail, if you would.

9 (Chu Deposition Exhibit No. 10 was
10 marked for identification.)

11 EXAMINATION

12 BY MS. BARNEY:

13 Q. Did anybody forward that e-mail to you do
14 you recall?

15 A. No.

16 Q. In your experience do you think if an
17 operator had concerns like these that he should
18 be sending them to his supervisor?

19 A. I think that's his job. Normally it
20 would not require an e-mail. You would normally
21 pick up the phone and go down the hole or
22 whatever and communicate this information.

23 Q. Did you come to learn in your time at
24 Burnside that Elizabeth Cromwell didn't always



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1 return phone calls?

2 A. Yes.

3 Q. So, if telephone wasn't the quickest
4 communication, then an e-mail in your experience
5 would have been appropriate?

6 A. Yes. An e-mail is a way to document that
7 a communication went through or there is not much
8 trust between two people.

9 Q. So, at this time period there was some I
10 guess issues about operators feeling like they
11 had to document things with management?

12 A. I'll say before that.

13 Q. Did you think there was some
14 justification in that with the operators'
15 concerns?

16 A. Yes.

17 Q. If there is some formal report that
18 DuPont would like to have filled out like an
19 initial incident report in your experience at Red
20 Lion or otherwise would it be a problem for the
21 operator to send an e-mail and do a formal report
22 if the supervisor wanted a formal report? Let me
23 -- I'll try to rephrase it.

24 There has been discussion about you



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1 have to do a first report or initial report and I
2 guess my question is is there anything wrong as
3 far as you know in your experience at DuPont with
4 sending an e-mail and doing a formal report?

5 A. I don't see it as wrong. It's unusual.

6 Q. You think going to the step of a formal
7 report right away would be unusual?

8 A. I mean, the incident report is normal.
9 An incident report is initiated shortly after the
10 incident is discovered and that document is left
11 open until all the available information is
12 gathered, analyzed and documented before
13 finalizing it. So, the incident report is a
14 formal report that is normal. Sending an e-mail
15 documenting operational maintenance issue is
16 unusual.

17 Q. You mean DuPont-wide it's a little
18 unusual to --

19 A. Well, I don't know about DuPont-wide. In
20 my experience it's unusual. I can't speak about
21 all DuPont. I look at least at this specific
22 topic on Exhibit 10, you know, is highly unusual
23 to have to document in e-mail. Now, e-mail is
24 not a formal document, but document in e-mail the



1 fact that an operator had communicated
2 information to his immediate supervision.

3 Q. And it was because of some distrust or
4 tension between management and operators?

5 A. Again, that would be my impression that
6 just between Jeff and Elizabeth, but, you know.

7 Q. So, at Burnside there was like you said
8 this justification for the operators to document
9 what they were saying to management from time to
10 time?

11 MS. WEINER: What she is asking for
12 is your opinion.

13 MS. BARNEY: Well, or perception I
14 guess.

15 A. Yes, I think they would be justified in
16 doing that, yes.

17 Q. Have you seen S03 gas leaks coming from
18 operating or process equipment before?

19 A. Yes.

20 Q. So, you kind of know how they look?

21 A. Right. Particularly since I now support
22 operations in China, Morocco, Chili.

23 Q. Let me see if I can show you a video on
24 here. I'm not sure how great this will be with



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1 my technology efforts. This should be -- I'm
2 trying to pull it up. There are several things
3 on the CD, which may make it a problem for
4 attaching it as an exhibit if I don't use all of
5 them.

6 You are looking at a video at this
7 time. This is actually a video of camera 13 at
8 DuPont Burnside, which I think is focused on the
9 converter. So, my question would be whether in
10 viewing this video you can see S03 leaking?

11 A. I'll say most likely, yes. This is a
12 converter and the top section is the 1st pass and
13 the exit of the 1st pass which, again, not
14 knowing the exact location, but that would
15 generally be where the 1st pass exit would have
16 S03, but I can tell you from -- so, S03 release
17 looks like a steam leak. So, visually I cannot
18 tell whether it's S03 or steam.

19 If I was there at the time the
20 circumstances I'm describing and the release is
21 exiting the 1st pass outlet, it can only be S03.
22 So, I don't know if that answers your question,
23 but visually, no, I don't have the eyes that can
24 discern molecular weight; but, if it's coming out



1 from the outlet duct of the 1st pass, the only
2 possible gas component that is released that will
3 have that kind of appearance is S03.

4 Q. And if Mr. Simoneaux testifies that he
5 took this video and that he saw that to be a S03
6 leak, would you have any reason to dispute that?

7 A. No. I know Jeff well enough that I would
8 not dispute that.

9 Q. And if Percy Bell thought it was S03, you
10 wouldn't dispute that either, right?

11 A. If Percy tells me that, I would not
12 dispute that at all because they will not just
13 show me the video, they will give me some
14 background information that would tell me -- from
15 this angle I can't tell. They would have told me
16 it's coming out of the duct because from this
17 angle you can imagine -- there isn't, but if
18 there are steam pipes leaking and they say what
19 is that, I wouldn't be able to tell. Percy or
20 Jeff, Percy in particular, I would not have any
21 problems trusting their information.

22 Q. Sometimes if there is a steam release
23 going on at the same time as an S03 you might be
24 able to differentiate the steam from the S03, but



1 if you are just looking at one cloud --

2 A. Yes.

3 Q. -- sometimes it's hard to tell, is that
4 right?

5 A. Right, right.

6 Q. Because steam will dissipate quicker and
7 it won't hang around as long?

8 A. And depending on the steam leak is a
9 little more energetic, right, because the steam
10 lines are a higher pressure and I don't know of
11 any steam pipes running that close to the
12 converter. Again, it's easy to check. I mean,
13 the operators have cameras especially at night to
14 be able to monitor the plant because usually it's
15 only two people there. These are cameras also
16 for safety. The field operators are there to
17 monitor what he is doing so that he is safe.

18 If they see something like that,
19 their first reaction is not assume it's a leak.
20 They see a leak and they send the operator to
21 verify where the leak is coming from.

22 Q. Mr. Chu, were you involved in any
23 particular discussions about shutting down the
24 plant to fix the leaking equipment?



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1 A. Specifically the CIP and HIP?

2 Q. Yes, or the converter.

3 A. Well, I was involved in requesting to
4 shut down the plant when they have converter leak
5 -- excuse me, converter waste heat boiler leaks.
6 That's when we have an internal leak of water
7 that is used to generate steam as a co-product of
8 the process and a couple occasions when they had
9 a boiler leak and it's leaking water internally
10 into the process, yeah, I told them they need to
11 shut down. Elizabeth and Gene were trying to
12 figure out if there is other explanation that
13 would result in what we were seeing of the
14 condition in the plant and I said, No, this
15 converter was boiling, you need to shut down now,
16 but not specifically on CIP and HIP.

17 Q. When was that that you had that
18 conversation about the boiler?

19 A. Probably 2010, towards the end of 2010 I
20 think.

21 Q. Was Tom Miller already the plant manager
22 at that point?

23 A. No. He was turnaround 2011.

24 Q. So, Don Janezic was the manager then?



1 A. Yes.

2 Q. Do you recall whether they shut down like
3 you recommended?

4 A. Yes.

5 Q. They did?

6 A. Yes.

7 Q. Before you transitioned out of the role
8 where you were involved with Burnside did you
9 have the occasion to recommend that they shut
10 down the plant before you left in order to make
11 any other repairs?

12 A. No, not that I can recall. I mean,
13 boiler leaks in my mind were major ones, so. The
14 reason I remember is I was having dinner in Baton
15 Rouge with my brother and I got called about a
16 boiler, potential boiler leak, that they just
17 repaired. So, I had to go back to the plant and
18 look at the information and say, yeah, it's a
19 boiler leak and shut down now. The dates, 2010,
20 2011, sometimes it's a blur those years, but I
21 was still the converter waste boiler leaks I
22 would have been responsible for them for the
23 plant.

24 Q. I'll show you one more video. I'm



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1 showing you a video that is dated October 27,
2 2013. And I ask you just to look at this one for
3 me. If you need to turn it, feel free. For the
4 record Mr. Simoneaux would testify that this was
5 taken by him outside on River Road or on the levy
6 or in the parking lot next door. I think he
7 moves around during his video.

8 A. Was this during the turnaround?

9 Q. I think the testimony has been that they
10 started back up after the shut down on this day,
11 on October 27th, that they just restarted. If
12 you look at marker one, colon 51, and stop it for
13 a second does it appear at that point in the
14 video where you can see a steam source and you
15 can see it behaving a little differently than
16 another gas source? Just tell me if you agree
17 with that.

18 A. That steam source dissipates easily. The
19 other one is something else. I mean, I can't
20 really -- all I can say is that it's unusual.

21 Q. It's a lot of -- whatever it is it's a
22 lot?

23 A. Right, I don't know what it is, but it's
24 unusual.



1 MS. WEINER: Would this be unusual if
2 they were just restarting plant operations that
3 he was videoing it?

4 THE WITNESS: Well, if they just
5 restarted it then the source of this kind of
6 fumes it would be at the stack level, not around
7 the plant.

8 EXAMINATION

9 BY MS. BARNEY:

10 Q. And you are seeing it around the plant?

11 A. I'm seeing it around the plant. Were
12 there other plant cameras that they were to
13 provide different angles?

14 Q. I think there are, I'm not sure how good
15 they depict.

16 A. Usually in the plant start-up because of
17 the type of plant Burnside is it starts up as a
18 sulfur burning plant they're initially putting
19 sulfur feed on. There will be a short period of
20 time where the stack, which is this one here --

21 Q. Let's stop it right here. We are talking
22 about marker 350 that is right above that tree?

23 A. Right, right. There may be an initial
24 plume and soon after the plant starts heating up



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1 and coming up to rates, it will dissipate; but,
2 coming out from the area that surrounds the
3 converter -- I can't tell if it's converter or
4 gas heat exchanger area. That is unusual.

5 Q. And you are referring to the area to the
6 left of the tree?

7 A. Right.

8 Q. And lower than the stack?

9 A. Right. I'll call that unusual and -- but
10 I can't determine what it is from just looking at
11 it.

12 Q. If Mr. Simoneaux testified that he
13 believed that to be SO₃ gas, would you have any
14 reason to dispute that if he is physically on the
15 parameter?

16 A. I can't dispute that he believes it. I'm
17 not saying that it's not, but, again, I can't --
18 I will say even if it's SO₃ I'm not sure where
19 it's coming from and how bad it is and why it's
20 allowed to continue if it is, in fact, SO₃
21 because that's a significant amount and almost at
22 ground level. Because that starts affecting
23 people that are in the general area as opposed to
24 dissipating at some altitude.



1 I will also say that I don't have a
2 lot of experience with big S03 leaks. It's not
3 something normal that is allowed in an acid
4 plant.

5 Q. So, if that is S03, that's disturbing to
6 you?

7 A. Yes. It would be, particularly if I have
8 to work there or live around there. I would say
9 with a release like that there would be notice to
10 some surrounding community complaints depending
11 on the wind direction. It's a different angle.

12 Q. If Mr. Simoneaux testified that he talked
13 with some people at the neighboring facility who
14 said their throat was irritated and eyes were
15 irritated, would that be consistent with the kind
16 of outcome you would expect with a leak like
17 that?

18 MS. WEINER: I just object to the
19 extent it calls for expertise outside of his
20 area, but you can answer it, if you know.

21 A. Yes, S03 would cause irritation to the
22 throat and eyes.

23 Q. It's not your testimony that this is not
24 S03 gas on the video, you are just not sure?



1 A. Yes, I'm just not sure. Even if it's
2 S03, suspected as S03, that would be pretty
3 disturbing to me. Can I ask what was the
4 duration of it?

5 Q. I believe -- well, I believe the evidence
6 is that the plant was shut -- rates were cut back
7 at some point and I believe the police were
8 called and the rates were cut back I believe is
9 the testimony that has been given.

10 MS. WEINER: But we don't have an
11 absolute -- I don't think we have a time that we
12 know at this point.

13 THE WITNESS: That the rates were
14 shut down?

15 MS. WEINER: I thought you were
16 asking the duration that looked like that and I
17 don't know that we have that in terms of hours.

18 THE WITNESS: But shortly after that
19 --

20 EXAMINATION

21 BY MS. BARNEY:

22 Q. The testimony is Mr. Simoneaux called the
23 police and that eventually a policeman came out
24 and eventually DuPont was called and after the



1 police called DuPont the rates were cut back at
2 the plant.

3 MS. WEINER: That's according to Mr.
4 Simoneaux.

5 MS. BARNEY: Yes.

6 A. After the rates cut down whether that got
7 reduced?

8 Q. There is a video for the next evening
9 that I might be able to show you.

10 MS. WEINER: I think the next video
11 was October 31st, right? The one we don't have.

12 MS. BARNEY: But you do have the
13 28th, next night.

14 THE WITNESS: That's why I mentioned
15 earlier that it's hard to determine how well the
16 plant is running just by looking at data in the
17 historian. Things like that is not recorded and
18 the historian to look back. That's why we have
19 operators, maintenance folks, to look.

20 EXAMINATION

21 BY MS. BARNEY:

22 Q. Whoever was there that night should have
23 been aware of that situation and done something
24 about it before a call from the police?



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1 A. Yes. Yes.

2 Q. I'll show you a video from October 28th.

3 MS. BARNEY: Which you do have,
4 Monique. The October 31st I don't have even have
5 off of the camera.

6 EXAMINATION

7 BY MS. BARNEY:

8 Q. I believe this is October 28th. Now you
9 are looking at a video from October 28, 2013.
10 You still see some white vapor, but does it
11 appear to be not as much?

12 A. It's hard to gauge, but regardless of
13 what I see in the video what does the plant
14 logbooks show us as an explanation for that
15 because that's not normal? I do agree with the
16 assessment that if it's just steam it dissipates
17 a lot quicker. That tends to linger. Whether
18 that's SO3 or smoke or something it's lingering a
19 lot longer than just steam vapor, but I would
20 expect that this -- again, at night it's more
21 visible because we have reflection of the lights
22 against the dark sky, but in the daytime that
23 would not be visible either.

24 Q. But the way that white fume material is



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1 looking is consistent with SO3 gas, but you are
2 not going to be in a position to say that it is
3 SO3 gas?

4 A. Right, regardless what it is if I'm at
5 the plant or anybody is at the plant there would
6 be some record or something that -- to explain
7 that. I mean, you know, dust particles can do
8 that too, but it's kind of disturbing. You can
9 see how this dissipates a lot quicker.

10 Q. You are pointing to the steam area.

11 A. Right.

12 Q. You can tell the difference between the
13 steam and the other material?

14 A. Yes. I guess I'm not the first -- well,
15 there has only been -- what is this, December?
16 I'm not the first one to see that in a
17 deposition?

18 MS. WEINER: No, you are not.

19 Q. Maybe October 28th you are, but not
20 October 27th.

21 A. And this is 2013, right?

22 MS. WEINER: Yes.

23 Q. Yes. And that was after a shutdown to
24 try to find the source of the leaks, which I



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1 believe you testified you heard from Percy Bell
2 that they weren't able to find all the leaks?

3 A. Yes.

4 Q. With the fuming looking appearance that
5 you saw on the October 27th video and the
6 October 28th video would you expect there to be a
7 full investigation of what that is and -- I'll
8 stop there. Would you expect there to be a full
9 investigation of what that is?

10 A. Yes.

11 Q. Would you expect as part of that that
12 they would run a calculation if they found a leak
13 source?

14 A. Yes.

15 Q. Would you expect that they would not
16 crank up again until they found the leak source
17 if there were a leak?

18 A. Would I expect that they would or
19 shouldn't?

20 Q. Would not crank back up or should not
21 crank back up.

22 A. I would expect that they should not. I
23 would expect they should not crank up unless they
24 can control it.



1 Q. I think we touched on this earlier, but
2 the hose and the box vacuum system -- let me
3 start over. The hose apparatus that Burnside has
4 attached to the leaking equipment that's not part
5 of the engineered design of how the plant is
6 supposed to operate, right?

7 A. That's correct. You are talking about
8 the black hose part, right?

9 Q. Right.

10 A. Part of the manifold system that is
11 connected to the dry tower to control fields from
12 pump tanks, which Burnside has, is part of the
13 system.

14 Q. The vacuum part?

15 A. Right. Again, the dry tower is a vessel
16 that has probably the lowest pressure, so, the
17 highest vacuum. So, it's used to -- manifolds
18 are designed to suck off gases from process
19 equipment. The corrugated hoses are not part of
20 the design.

21 Q. That apparatus is just sort of attached
22 to a vacuum system?

23 A. Right.

24 Q. And the vacuum system has an engineered



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1 design purpose and it is kind of borrowed and
2 attached to the hose material, which the hose
3 material is not part of the engineered design?

4 A. Right, the hose's specific purpose, let's
5 say, in this application is to control the gas
6 leak and plants are not -- even though they
7 expect to have gas leaks you don't design a
8 system to -- just for the gas leaks. It was
9 extended, that system was extended towards this
10 application.

11 Q. During your involvement with the Burnside
12 site did you ever become aware that DuPont had
13 reported any of the gas leaks from the CIP, the
14 HIP, the converter the super heater or any of
15 their duct work to EPA or DEQ?

16 A. Mainly hearsay. Again, I knew they were
17 supposed to be reporting it. Do I look at the
18 reports? No. No.

19 Q. If the testimony has been that they have
20 not been reporting it, do you think one reason
21 for that is that they don't want to pay the fines
22 that might go along with reporting it?

23 MS. WEINER: I'll object to the
24 extent it calls for speculation. You can answer



1 if you know.

2 A. I don't know.

3 Q. Wasn't it your impression when you were
4 down at Burnside that that was part of the
5 motivation behind not reporting?

6 MS. WEINER: Same objection.

7 A. Again, whether reporting or not for
8 whatever reason, whether it's fines, sometimes it
9 would be -- the reason may be as simple as not
10 making the plant look bad by having this many
11 problems, right? Because an incident report is a
12 document and it's part of a benchmark to compare
13 other plants and, of course, the plants that have
14 least amount of reports look better. Again, a
15 speculation as to the reason why they would do --
16 whether they report or not.

17 Q. If during a dinner conversation back in
18 spring 2012 you mentioned that DuPont wasn't
19 reporting these leaks because they didn't want to
20 pay a fine of \$25,000.00 a day, could it be that
21 you just don't remember saying that?

22 A. It could be I couldn't remember saying
23 that.

24 Q. But you may have said that?



1 A. I may have said that.

2 Q. If any operators at the plant remember
3 that, you wouldn't dispute it?

4 A. Depends on which operator tells me that.
5 Even if I said it it would be a speculation at
6 that time as to why. I don't attempt to
7 understand the mentality of management.

8 Q. It may have been just a personal
9 impression that you got?

10 A. Yeah. It all depends if I knew that it
11 was reportable. Based on my calculation, my
12 estimate, it would have been incumbent on me to
13 report it, not necessarily to the EPA, but to my
14 management, which is now I don't report to the
15 plant.

16 Q. But you were never given the size of the
17 holes and cracks that were in these vessels for
18 you to make that determination, right?

19 A. Right, right. I can determine all the
20 process parameters, but the key parameters that's
21 a whole different issue that I believe can be
22 arrived at only based on discussions and a
23 consensus.

24 Q. So, if the operators that were at that



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1 dinner included Mr. Rapp or Mr. Simoneaux, would
2 you dispute the accuracy of their recollections
3 what you said about the fines?

4 A. Again, I wouldn't dispute it, but I
5 really don't recollect specifically saying it's
6 because of a \$25,000.00 fine. It might be a
7 statement that I made and they recollect
8 correctly, but I also don't find any reason for
9 them to make that up.

10 Q. I guess in your experience there could be
11 fines if you release S03?

12 A. Well, this reportable quantity that is
13 associated -- again, not just necessarily fines,
14 but there is internal ratings, classifications of
15 environmental events, doesn't reflect very well
16 on the plant management if they have categories
17 of release that are reportable.

18 Q. Are you familiar with TSCA, that
19 particular environmental statute?

20 A. I heard of it. I mean, I'm aware, yes.

21 Q. Do you know whether TSCA has a quantity
22 that triggers obligations under TSCA?

23 A. I know that there are quantities that
24 trigger. I don't know exactly what the



1 quantities are for specific compounds. I guess
2 you are referring to acid mist and SO3 or SO2.

3 Q. I guess my question is I realize there
4 are reportable quantities under various
5 environmental statutes, but do you know for sure
6 whether -- maybe it's a legal question, but do
7 you have an understanding as to whether there is
8 a specific reportable quantity under the TSCA
9 statute?

10 A. I know there is specific quantity. I
11 don't know what the quantity is.

12 MS. WEINER: You mean you know there
13 is a specific quantity, but you don't know what
14 it is?

15 THE WITNESS: What the number is.

16 EXAMINATION

17 BY MS. BARNEY:

18 Q. And you think that that quantity
19 correlates to the specific legal statute for TSCA
20 notification?

21 A. Yes.

22 Q. That's how you -- whatever training you
23 have had on environmental issues that's your
24 understanding that there is something in the TSCA



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1 statute that has a quantity?

2 A. Mm-hmm, yes, either quantity or
3 concentration.

4 Q. Other than your college education and
5 your job at Mittelhauser have you had any
6 environmental experience outside of working at
7 DuPont?

8 A. Even at Mittelhauser my role was not on
9 environmental.

10 Q. That's true, let me strike that and start
11 over. Have you had any environmental education
12 or training outside of DuPont?

13 A. Hazardous waste classification. I mean,
14 that would be considered environmental, yes.

15 Q. Where did you have that training? At
16 Conoco?

17 A. I think probably Mittelhauser and also
18 Conoco because we had to deal with hazardous
19 waste management for exploration and production.

20 Q. Can you think of any other environmental
21 training that you have had other than at DuPont?

22 A. Not that I recall.

23 Q. In any of your experience at DuPont have
24 you ever heard a plant manager discouraging



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1 employees from calling authorities about gas
2 leaks or any other environmental problem?

3 A. Yes, Burnside.

4 Q. Tell me about that.

5 A. Well, again, this is from what the
6 operators will tell me, not as an official policy
7 that the plant manager would go out and say don't
8 call.

9 Q. You heard from the operators at Burnside
10 that the plant manager was discouraging people
11 from calling the outside authorities?

12 A. Yes.

13 Q. I guess you didn't have any involvement
14 or did you with T. J. Ozbun at the Burnside site?

15 A. What do you mean by involvement?

16 Q. He is now the environmental coordinator,
17 but I don't think he had that role when you were
18 there, did he?

19 A. No. We have offices in the same trailer,
20 but not really any day-to-day interactions.

21 Q. I guess you did have interaction with
22 Kerry Long when he was the environmental person,
23 we talked about him.

24 A. Mm-hmm.



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1 Q. We talked about the time he asked you for
2 advice to make sure he had the parameters for
3 some of those calculations. Is there any other
4 interaction that you recall having with Mr. Long?

5 A. Yeah, in one of the exhibits, which was
6 to estimate acid that leak out of the pump tank.

7 Q. That was your Exhibit 2 that I think you
8 brought with you, is that right?

9 MS. WEINER: Right.

10 Q. Here you go.

11 A. Right. I was asked by somebody in
12 Wilmington to check the original estimate for the
13 acid leak as a result of -- that resulted in a pH
14 excretion into the river. So, I had to ask Kerry
15 how they estimated it and then I revised their
16 estimate based on the DCS data and talking to --
17 with the people that were at the site that had
18 knowledge of the incident. But, again, I wasn't
19 part of the incident investigation. Just as I'll
20 say an additional source to recalculate what the
21 range of acid that could have leaked into the
22 river.

23 Q. Before you got involved -- well, strike
24 that. Who had done the first estimate?



1 A. Kerry.

2 Q. And, so, somebody at corporate asked you
3 to double-check the estimate?

4 A. Right.

5 Q. Who at corporate asked you to
6 double-check?

7 A. Maureen Miller.

8 Q. What is her role?

9 A. I forget what her current role is. At
10 that time she was safety, health and environment
11 manager I think for the acid circuit, but she
12 also the first time I met her years back she was
13 an environmental specialist at Burnside.

14 Q. When you double-checked the estimate, did
15 you come up with a higher range or a lower range
16 than the Burnside folks, Mr. Long, had come up
17 with?

18 A. A higher range.

19 Q. Did Mr. Long tell you how he arrived at
20 his estimate?

21 A. Yes.

22 Q. And how had he done that?

23 A. He looked at the ground and came up with
24 a number, which is the reason I was asked to look



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1 into it.

2 Q. Did Mr. Long give any explanation for why
3 he just looked at the ground and did a number?

4 A. I did not ask.

5 Q. Did you get the impression in dealing
6 with Mr. Long that he was under some pressure to
7 keep the numbers low?

8 A. No.

9 Q. You didn't receive that impression?

10 A. From talking to him, no.

11 Q. Do you know why he left DuPont Burnside?

12 A. No, I don't know. I think he went to a
13 different company or something, but, again, I
14 have no animosity. We just didn't interact that
15 much. He is not a very sociable person, so, you
16 know, I just wasn't interested.

17 Q. Do you know whether Maureen Miller has
18 any information about the hose system or the
19 leaks at the Burnside plant?

20 A. I don't know, yeah, I wouldn't know.

21 Q. When you were involved with Burnside, was
22 there a person, Matt Barnes, who was in the
23 environmental group?

24 A. Yes.



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1 Q. What was his role in connection with
2 Burnside? I know he wasn't at Burnside.

3 A. I think he took over Maureen's role.
4 Matt used to work for Maureen. Again, the titles
5 over the years escape me. I knew at one point I
6 first met Matt at Red Lion.

7 Q. When is the first time you met Tom
8 Miller?

9 A. Turnaround of 2011, April. Could be
10 March.

11 Q. The calculations that we have talked
12 about earlier, Exhibit -- I won't repeat the
13 numbers, I'll never get it right -- the
14 calculation formulas that we were talking about,
15 those would not measure the amount of gas in the
16 hose system, right?

17 A. You mean the hose that is used to --

18 Q. Try to suck up the leaks.

19 A. No. It measures the gas that has
20 potentially been sucked into the hose. So, the
21 calculations are designed to see what is leaked
22 out of the hole.

23 Q. Out of the hole?

24 A. Yes. It doesn't calculate the efficiency



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1 by which the hose is sucking or containing the
2 leak.

3 Q. If the hose is capturing some of the gas
4 if the hose melts down, then that gas escapes,
5 right?

6 A. No. If the hose melts, then it reduces
7 the efficiency of the suction. So, whatever is
8 in the hose is already gone into the system, into
9 the dry tower. So, if it melts now you have a
10 larger opening, so, the hose system is trying to
11 suck more air out of a bigger opening.

12 Therefore, it reduces the effectiveness of trying
13 to localize that hole right at the leak.

14 Q. So, more of the gas may get away and not
15 be sucked up by the hose?

16 A. Right.

17 Q. When the hoses melt, is that sometimes a
18 product of acid building up in the hose?

19 A. That is one component. The other
20 component is depending on where the leak is some
21 of those gases exiting could be 400 to
22 600 degrees C. So, part of the reason it's
23 melting is being it's sucking fairly hot gas.
24 So, melting by itself depending on the type it



1 could be the temperature of the gas itself or the
2 fact that acid is being formed because SO₃ is
3 reacting with water and releasing more energy
4 localizing the heating even more.

5 Q. When the acid builds up that affects the
6 vacuum, right, are you familiar with that, that
7 if acid fills up in a section of the hose, then
8 they lose vacuum or are you not familiar enough
9 with that apparatus to know?

10 A. Well, the hose filling up with acid or
11 any kind of liquid would not affect the vacuum.
12 It may affect the effectiveness of the suction.

13 Q. Okay, I worded that wrong.

14 A. But I understood I think what you are
15 driving at. Any way that restricts the hose,
16 position, size, will affect the effectiveness of
17 trying to control the gas leak.

18 Q. And depending on how effective the vacuum
19 is being when the hose melts would determine how
20 much gas is in the hose and released when it
21 melts, is that right?

22 A. Like I said, essentially once the gas is
23 in the hose -- let's say you have gas in the hose
24 and all of a sudden the bottom section primarily



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1 is going to be at the inlet, not in the middle,
2 melts, whatever gas is in the hose is going to be
3 sucked. Now, the new gas that is sucked that
4 concentration will vary. The effectiveness at a
5 time after it melts is compromised, but once it's
6 in the hose unless the hose ruptures before it
7 goes into the manifold, before it goes into the
8 dry tower, once it's in the hose --

9 Q. It should go back into the process?

10 A. It should go into the process.

11 Q. But the leak source at that point is not
12 contained as much as --

13 A. Contained, right, right.

14 Q. And these connections between where the
15 hose is positioned at the leak those are not
16 airtight connections, right, as far as you know?

17 A. There are two types. Ones that they can
18 put the hose right next to an area and other
19 times they try to build a box, let's say, a hood
20 by which most of the gas is contained in that box
21 and they use the hose to suck it. I don't know
22 which method they used.

23 Q. If the contractor who is trying to
24 capture the gas with this hose system says that



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1 it doesn't get all the gas, would you disagree
2 with him?

3 A. I would not disagree with that statement.

4 MS. BARNEY: I think that's all I
5 have.

6 MS. WEINER: I'm going to ask just a
7 few questions just to clarify a few things, but I
8 think we are getting close to being done.

9 EXAMINATION

10 BY MS. WEINER:

11 Q. Getting back to the questions that Ms.
12 Barney asked you about the plant reporting any of
13 the leaks from the CIP or the HIP or the
14 converter you said that you knew they were
15 supposed to be reporting. Did you mean reporting
16 to outside regulatory agencies or reporting
17 internally within DuPont?

18 A. Reporting internally with DuPont. That's
19 the primary criteria, you report internally and
20 based on that determine whether it's reported
21 externally.

22 Q. As you sit here today do you have any
23 information based on what you have seen, what you
24 talked to operators about or calculations that



1 you believe Burnside has exceeded reportable
2 quantities for S03 such that they should have
3 reported to regulatory agencies?

4 A. I have no information to say that they
5 have not reported what is supposed to be
6 reported.

7 Q. Ms. Barney asked you after you watched
8 that first video, the 10/27 video, which was the
9 one when the plant had started up that day, and
10 she asked you if something should have been done
11 about that before the police were called,
12 something done about that at the plant before the
13 police were called by Mr. Simoneaux. You said,
14 Yes, I would think so. That would be assuming
15 that it was an S03 release, right?

16 A. Right. And, again, it's also assuming
17 that they probably were doing something to
18 control it. From the video you can't tell that
19 nothing was done.

20 Q. Other than seeing these videos have you
21 ever seen personally an S03 leak at the Burnside
22 plant from the CIP, the HIP or the converter?

23 A. Yes, but not at this magnitude. I've
24 seen small leaks that at the time appeared to be



1 controlled by the hose, but I've seen fairly
2 large leaks outside the plant. So, I have a
3 gauge as to what leaks are supposed to look like
4 and not supposed to happen in acid plants.

5 Q. The leaks that you have seen at the
6 Burnside plant those were not ones that you
7 thought required any reporting to outside
8 regulatory agencies?

9 A. I have no way to determine that. I see
10 leaks, leaks are supposed to be, again,
11 internally documented and then determined whether
12 they're reportable or not.

13 Q. And you don't make that determination as
14 to whether they're reportable or not?

15 A. I don't make that determination. But I
16 would add that if you need a hose system to
17 contain it it's because there is enough that
18 could be calculated and then from that determine
19 whether it's reportable or not. You just don't
20 put a hose to every single leak that you see.

21 Q. While you were assigned to Burnside from
22 first quarter 2010 to first quarter 2012, that
23 two-year period, were you aware that the plant
24 had ever done any shutdowns to fix some of those



1 holes or cracks or leaks that were causing the
2 emissions?

3 A. Yes.

4 Q. Do you know if they were successful in
5 fixing some of the leaks and cracks?

6 A. They were successful in fixing some of
7 the leaks.

8 Q. And then potentially others would develop
9 so they may have continued to have problems, but
10 they did go in and fix some of them?

11 A. Again, I don't know the number. That
12 would be something that would be documented by
13 maintenance records that there will be leaks that
14 are successfully repaired, but they could develop
15 later on in the process and they have to fix
16 again. There would be leaks that they were
17 unsuccessful to repair the first, the second
18 time, and they tried to contain it and then
19 repaired at a later time. So, it's a number.
20 The success rates I don't know.

21 MS. WEINER: I think that's all I
22 have. Thank you.

23 MS. BARNEY: Just one follow-up.

24



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EXAMINATION

BY MS. BARNEY:

Q. You haven't seen enough documentation on the size of the holes and cracks and the leaking equipment to support any kind of calculations by DuPont as to whether the leaks have been reportable quantities, right? I'll just strike that. I think what I'm trying to ask you is that without having -- your testimony or I think everybody is understanding is they still haven't found all the leaks. So, without knowing the size and number of the holes and cracks in the leaking equipment as far as you can tell there is not the basic information that DuPont would need in order to know how much gas is leaking in order to quantify it, right?

A. Correct. Just like -- again, the calculation establishes the potential leak based on the information. How much is contained by the hose, how much escapes, how much is unaccountable for, again, is very uncertain.

Q. When you were out at Burnside, were you out at night a lot at the plant?

A. Generally, if I come at night and I do



1 visit the operators at night I will go out to
2 look for -- to check some processes, the towers;
3 but, generally when I go out at night, it's just
4 to visit with the operators. Now, if they
5 invariably ask me to go out to look at something,
6 and I can't remember a specific situation, they
7 want me to check on something, I usually go with
8 them. Again, that's what I like to establish
9 with the operators in the field that I tell them
10 if you see something that is unusual, let me know
11 because there is things that I cannot see by just
12 looking at a computer screen or from a thousand
13 miles away.

14 MS. BARNEY: I think that's all I
15 have. Please attach the October 28th video as
16 Chu 11.

17 (Chu Deposition Exhibit No. 11 was
18 marked for identification.)

19 (The deposition was concluded at 4:48
20 p.m.)
21
22
23
24



I N D E X

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E X H I B I T S

CHU DEPOSITION EXHIBITS	MARKED
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Exhibit 1	One sheet of calculations prepared by Luis Chu	56
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5 CERTIFICATE OF REPORTER

6 I, Christina M. Vitale, Certified Court
7 Reporter and Notary Public, do hereby certify
8 that there came before me on Friday, December 6,
9 2013, the deponent herein, LUIS CHU, who was duly
10 sworn by me and thereafter examined by counsel
11 for the respective parties; that the questions
12 asked of said deponent and the answers given were
13 taken down by me in Stenotype notes and
14 thereafter transcribed by use of computer-aided
15 transcription and computer printer under my
16 direction.

17 I further certify that the foregoing is a
18 true and correct transcript of the testimony
19 given at said examination of said witness.

20 I further certify that I am not counsel,
21 attorney, or relative of either party, or
22 otherwise interested in the event of this suit.

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